

Chalcogenide Glass for Active and Passive Mid-IR Applications

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Outline

- Background
- Purification and synthesis
- Optical fibre development
- Active devices in the mid-IR
- Concluding remarks

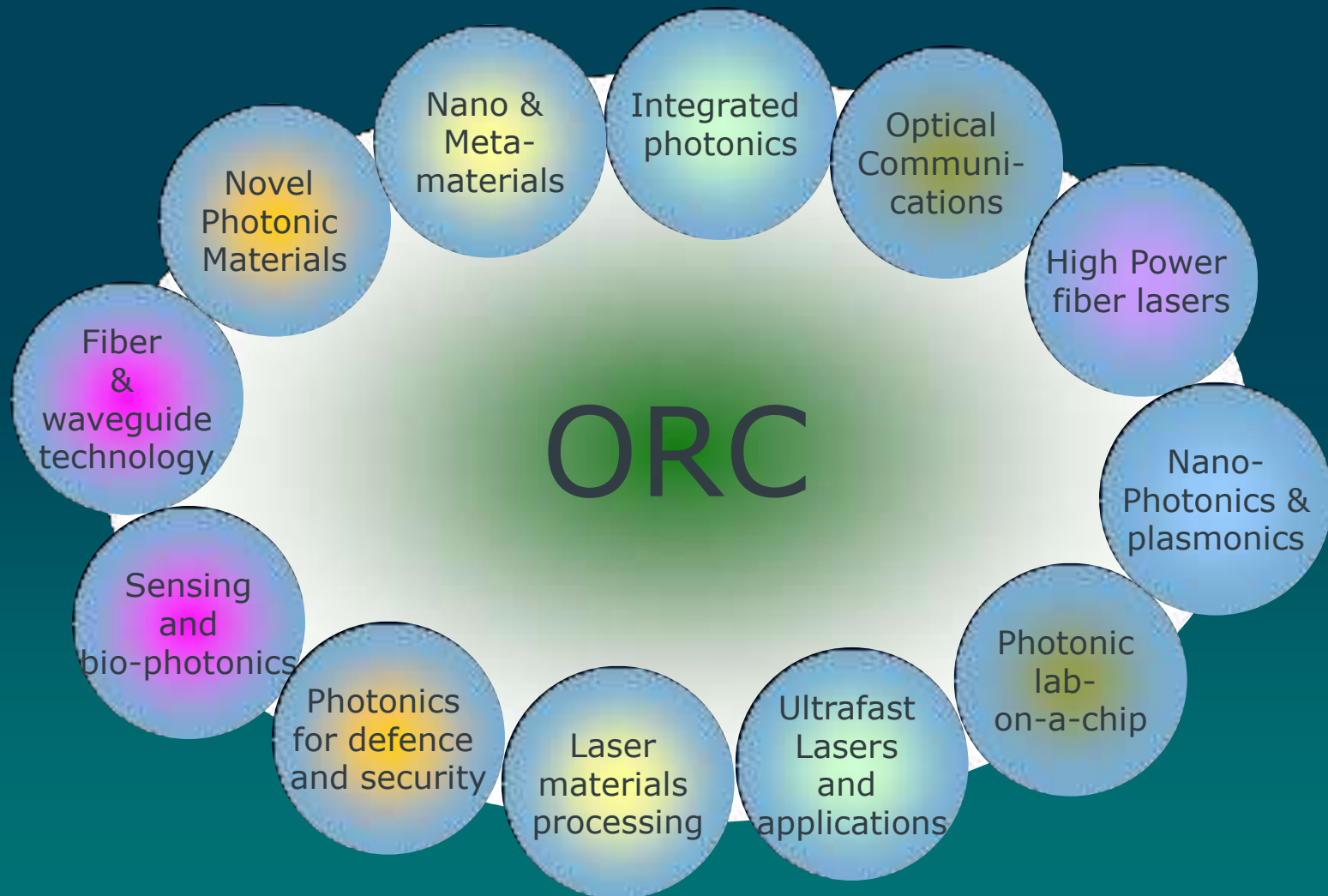


Optoelectronics Research Centre

- 40 year history beginning from ground breaking work in optical fibers
- Now the largest group in UK (170 staff / PhD students, 65 labs)
- Generates ~50% of our Universities Intellectual Property
- Extensive international industrial and University links
- A worldwide alumni of 600 staff many in senior positions
- A photonics cluster of 11 companies
- 270 Publications/11 Patents per year
- 50 Invited / Plenary talks per year
- Staff includes 3 Fellows of the Royal Society

Primary Research Areas

UNIVERSITY OF
Southampton
Optoelectronics
Research Centre



Purification & Synthesis



- Raw materials
- Reactive gas conversion
- Chemical vapour deposition



The Chalcogenides

What is a Chalcogenide?

- From Greek *sulphur-loving* for elements that frequently bond to sulphur
- Seen in various forms: crystalline, single crystal, quantum dots, phosphors, ceramics

Typical Amorphous Compositions

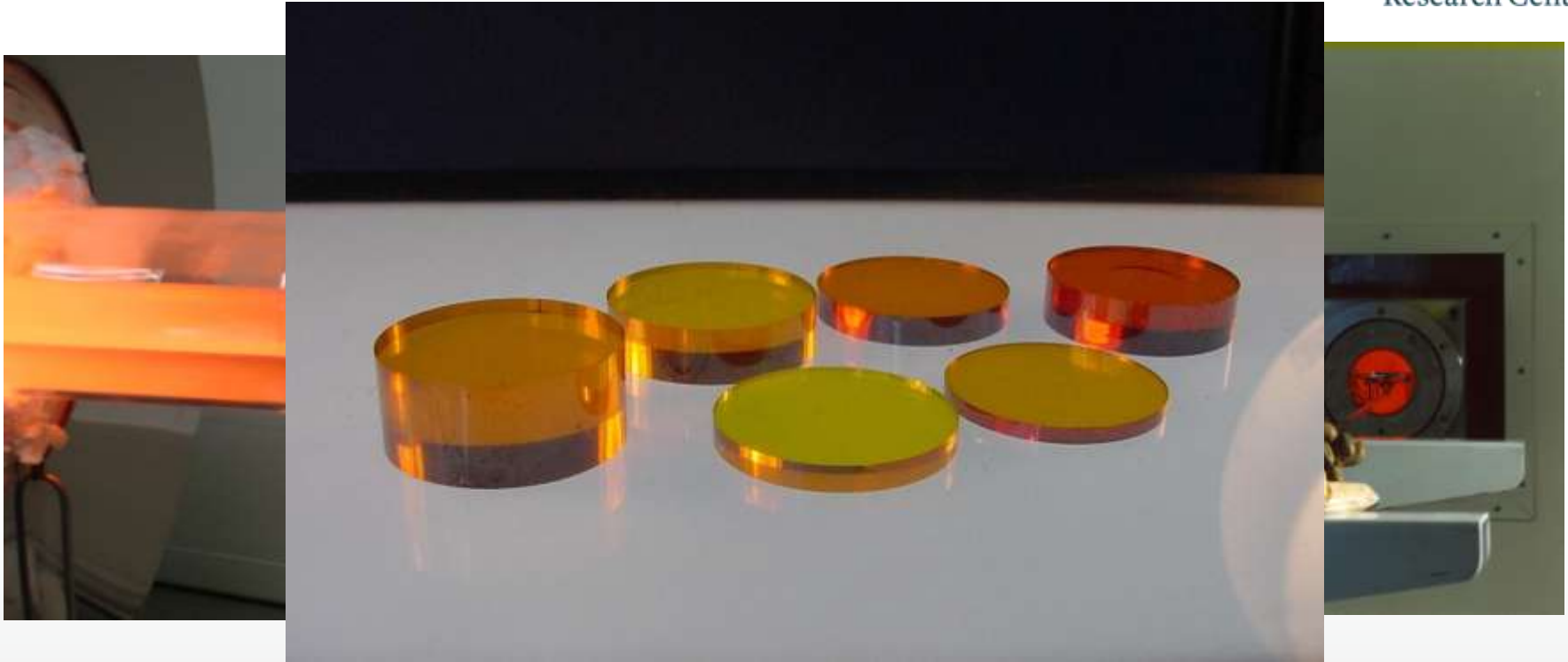
- As-S, As-S-Se, Ge-Sb-Te
- Predominately As or Se based (toxic!)

ORC Research Focussed On

- Gallium Lanthanum Sulphides (non-toxic)
- Germanium Sulphides (non-toxic)
- Capability to melt any glass composition exists

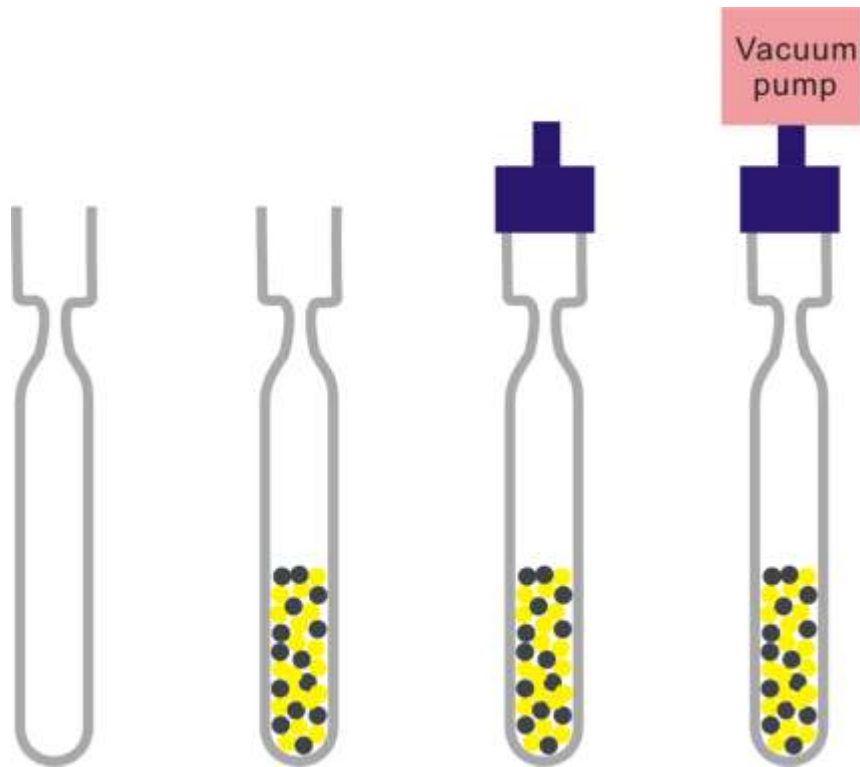


Glass Melting (Open Atmosphere)



- Wide range of horizontal and vertical tube furnaces, chamber furnaces, high and low temperature ovens, vacuum processing
- Processing in dry nitrogen, argon, oxygen, SF_6 , hydrogen and hydrogen sulphide
- Speciality heating including rapid thermal annealing and RF induction

Glass Melting – Sealed Ampoule Melting



Typically used for compounding elements, eg. Ge, Se, Te, Sb₈

RF Induction Heating

- Clean, precise, controllable heating
- Custom design, interchangeable coils
- Flexible, we configure the furnace to our needs



CIH_ISM HF20

Up to 20 kW output
50-150kHz

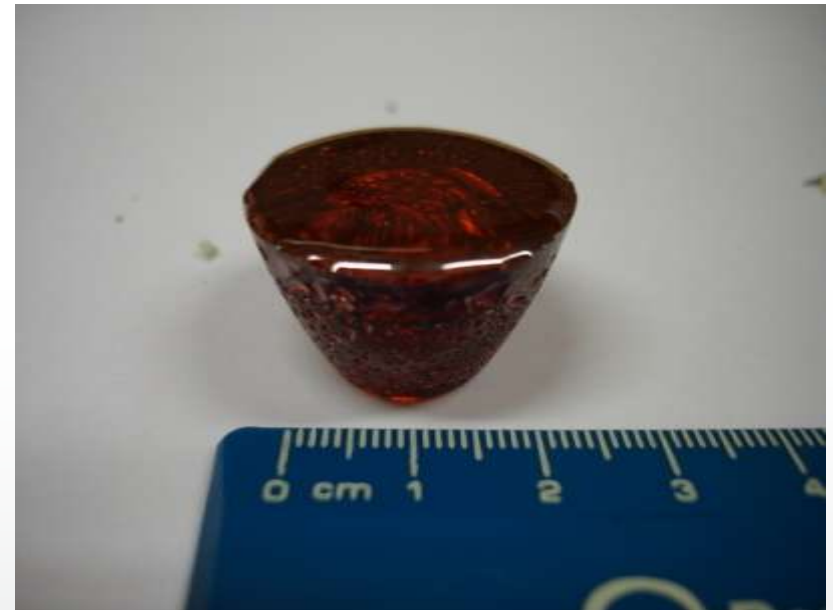


How Not to Melt Glass



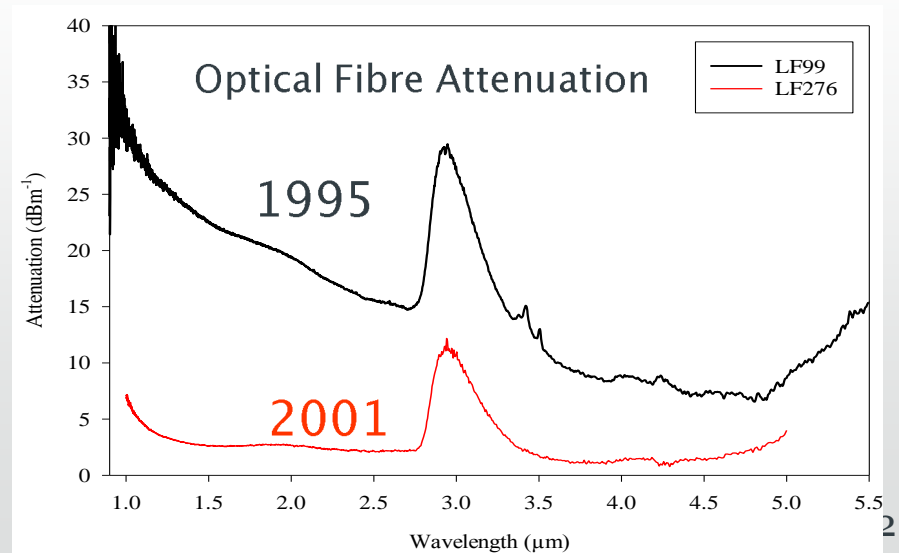
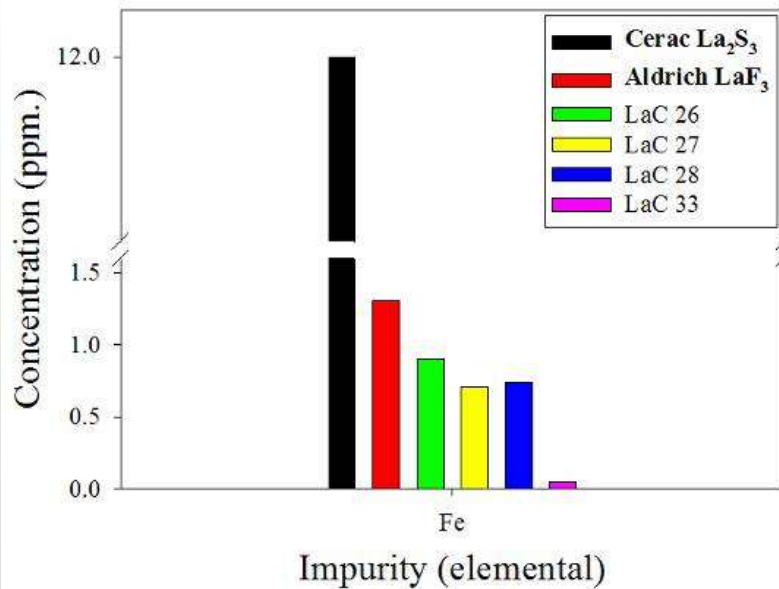
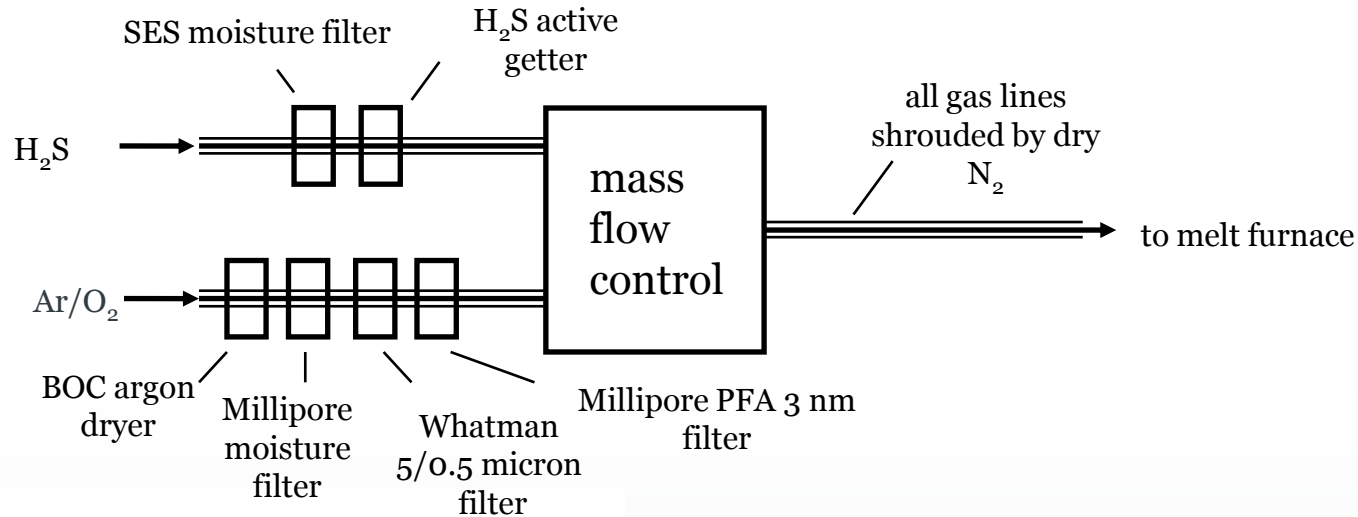
30 October 2005

Raw Materials are Critical!



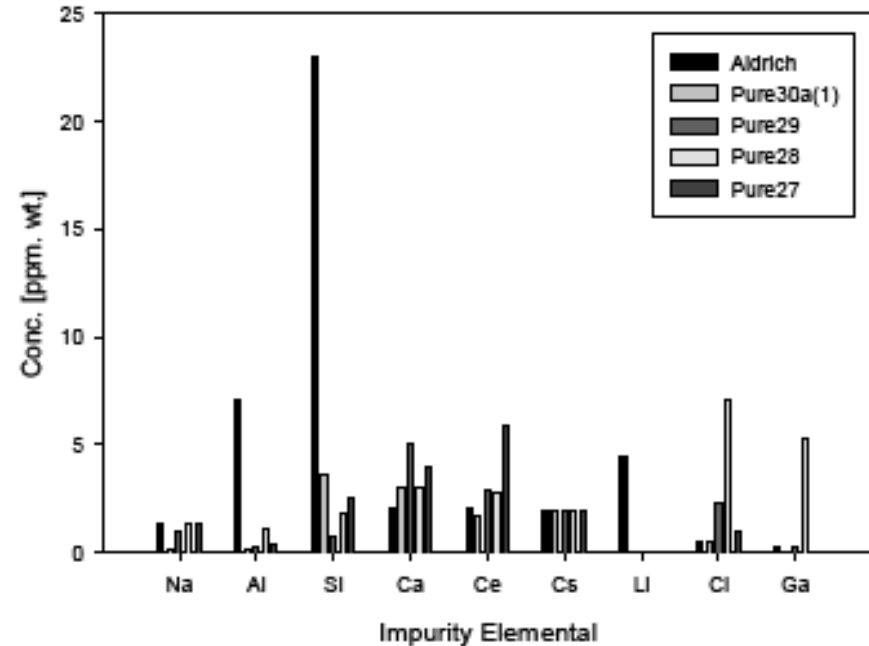
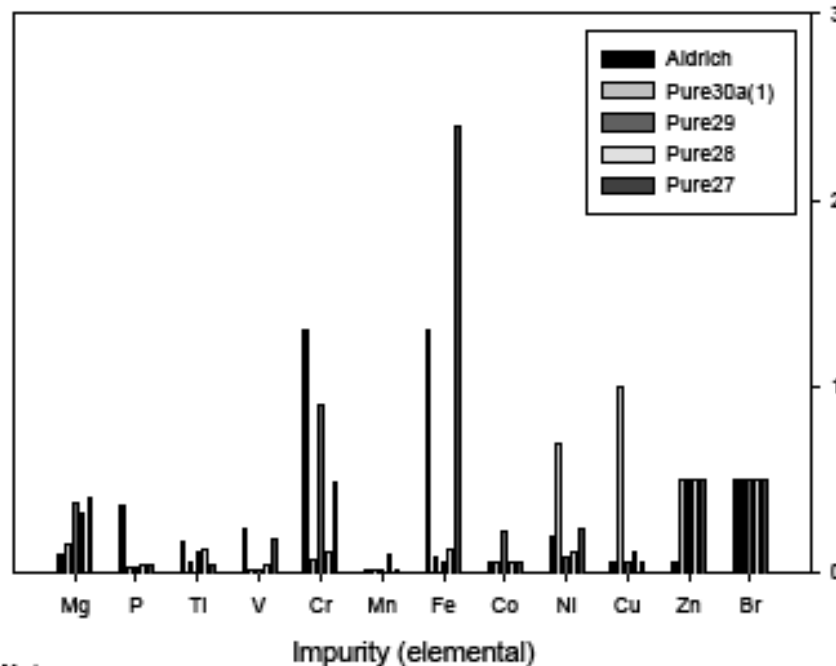
- Ga-La-S-O samples prepared with materials sourced from different manufacturers.
- Melting conditions were identical.

In-House Designed & Built Gas Delivery



Purity Levels for LaF₃

Aldrich vs in-house purified



Notes

Pure30a(1)

Tube + crucibles baked out = 24hrs. Crucible Type = Ceramic x 2
Tube clean (though used in Pure 30 / LaCO13 / LaCO14). Powder Mass ~ 8g / 8g

Transfer and use of pod proved impossible.
Crucibles exposed to atmosphere for 1min. Slight black colouring on surface after purification.

Pure29

Crucible Type = Carbon x 2. Powder Mass ~ 20g / 65g

Pure28

Tube + crucibles baked out = 12hrs. New Tube used.
Crucible Type = Carbon. Powder Mass ~ 83g
Appearance of powder after purification = white

Pure27

Tube + crucibles baked out = 12hrs. New Tube used with carbon liner.
Crucible Type = Carbon. Powder Mass ~ 60 / 30g

Purification Run	C	O	S
	(Impurity Content in ppm wt)		
Aldrich	140	100	0.71
Pure30a (1)	70	6000	140
Pure29	43	300	-
Pure28	45	570	160
Pure27	65	210	780

Thermal Properties

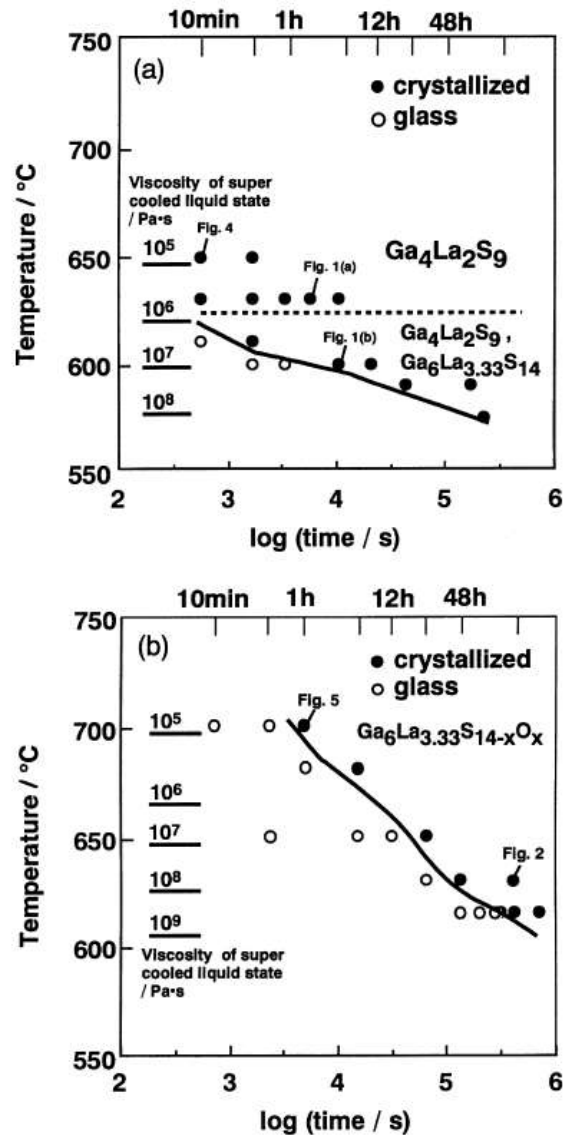


Fig. 3. TTT diagrams of (a) GLS and (b) GLSO glasses during isothermal treatments. Viscosity data of super cooled liquid states is also shown in these figures.

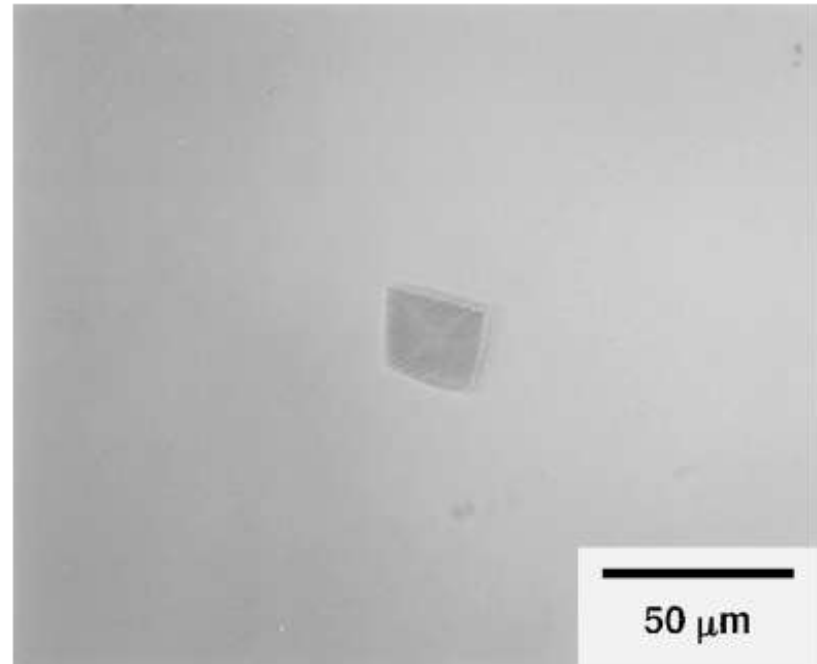


Fig. 5. Transmission optical micrograph of the GLSO glass treated at 700°C for 1 h.

Glass Modification

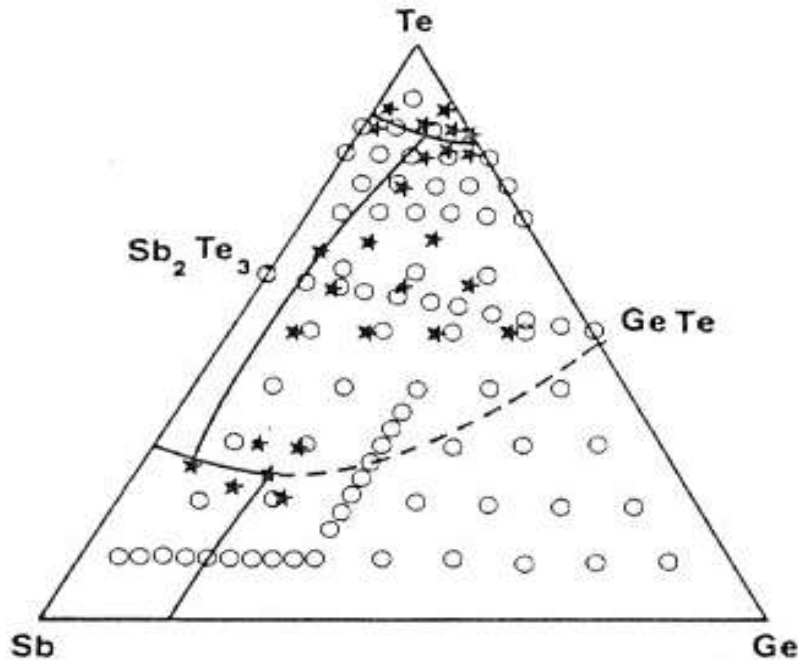
○ ORC Research 1992-2005

H																	He
Li	Be											B	C	N	O	F	Ne
Na	Mg											Al	Si	P	S	Cl	Ar
K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr
Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe
Cs	Ba	La	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rn
Fr	Ra	Ac	Rf	Db	Sg	Bh	Hs	Mt									

Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu
Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr

1 st Generation US 4115872 ~1978	2 nd Generation US 45335219 ~1994	3 rd Generation US 5341328 ~1994	4 th Generation US 5406509 ~1995	5 th Generation US 6011757 ~2000	Emerging US 4115872 ~2003
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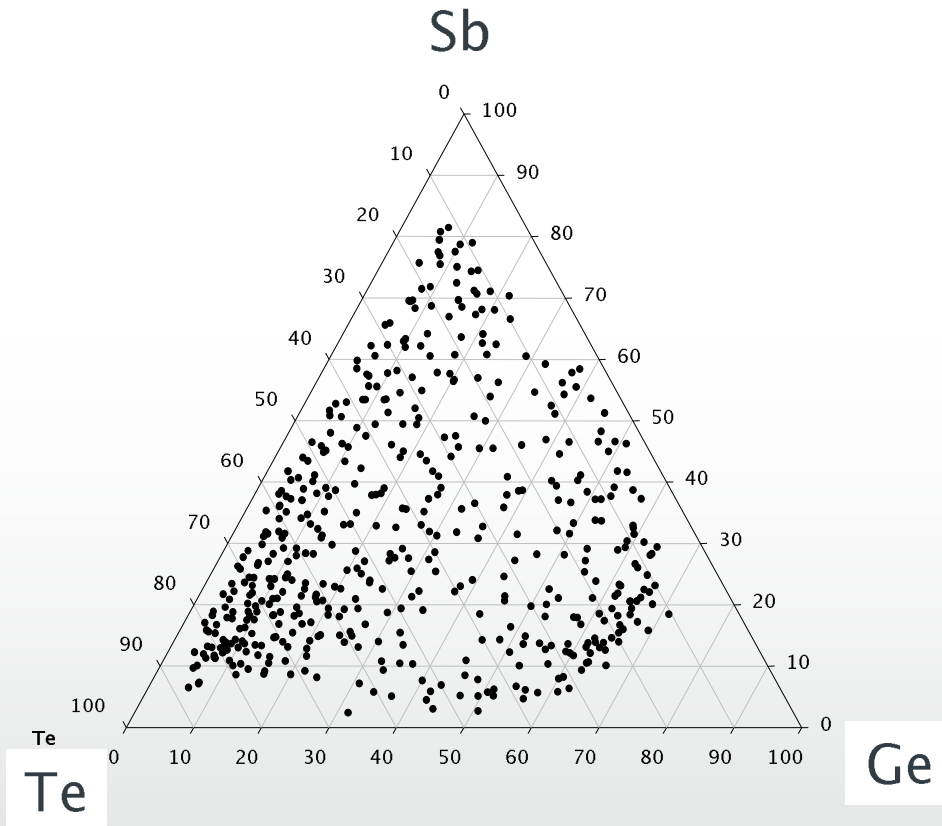
Background – Early Mapping of GeSbTe



- 100 samples GST alloys studied
- Each melted in sealed ampoules
- Composition individually analyzed
- Melting temperature by DTA
- Tg and crystallization by DSC
- Crystalline phases by XRD

Time Scale: several months?

Full Ternary Analysis



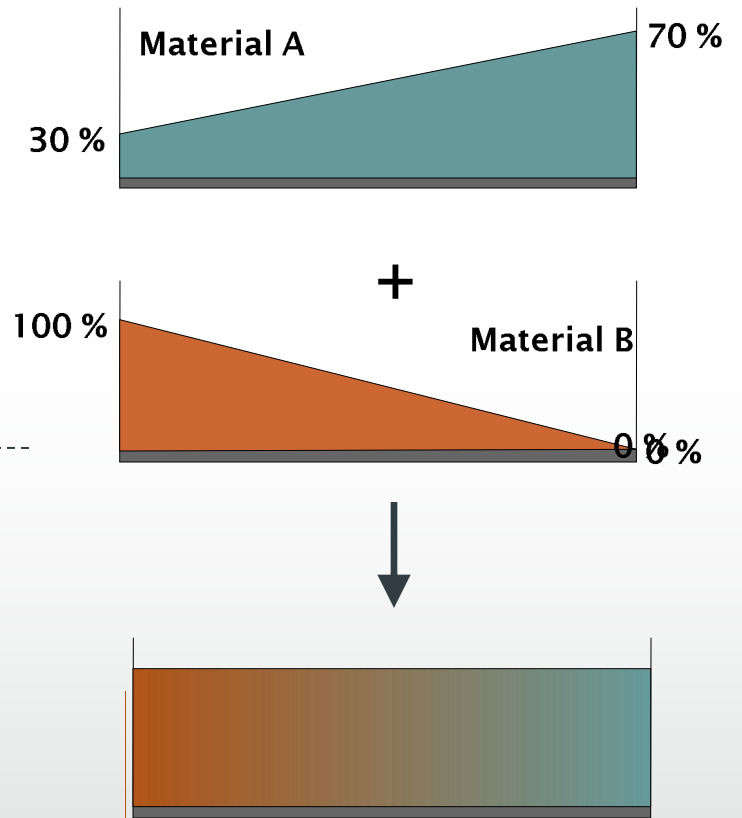
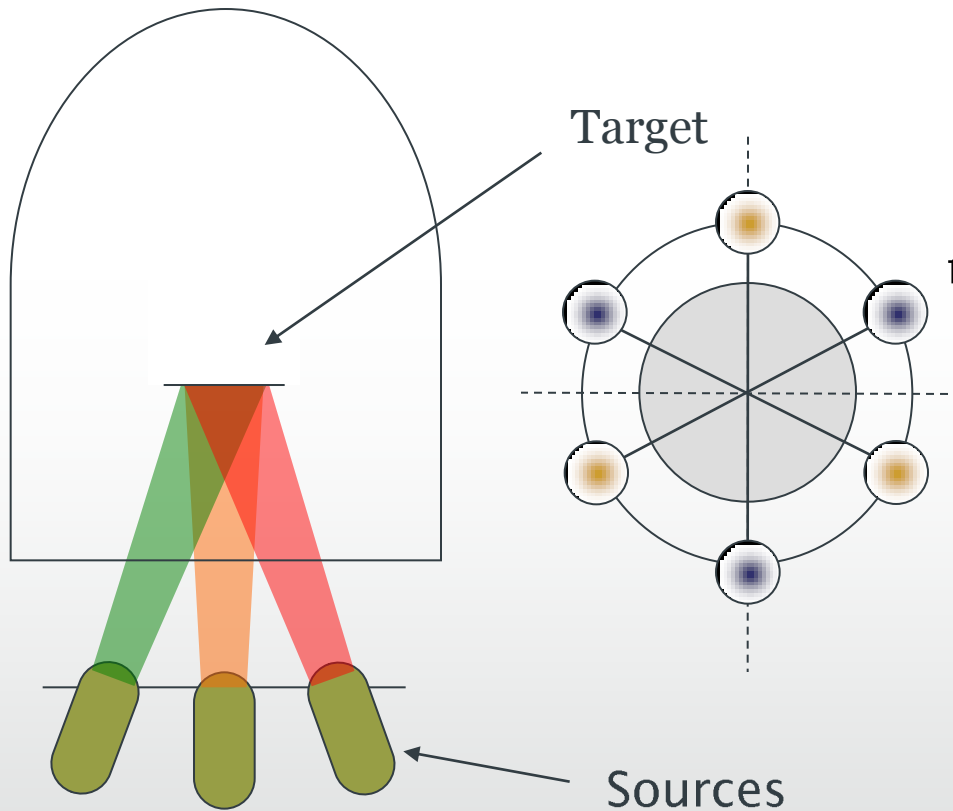
Calibration Runs:	2 - 3 days
Run 653	2 hours
Run 659	2 hours
Run 770	2 hours
Primary Screening	2 - 3 days

Time Scale: one week

Presented at *EPCOS '05* Cambridge Sep 2005
 R.E.Simpson, D.W.Hewak, S.Guerin, B.Hayden,
 G.Purdy, "High throughput synthesis and screening of
 chalcogenide materials for data storage"

Pioneering Technology: High Throughput Physical Vapour Deposition
 Material Discovery Times accelerated by a factor of 10 - 100

High Throughput Deposition

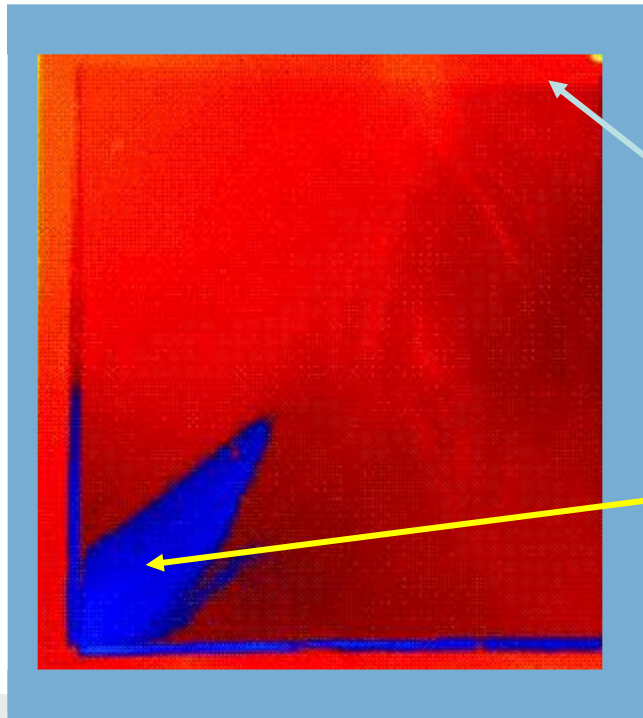


Shutter over each target ensures reproducible and reliable “wedge” which combined gives a desired gradient

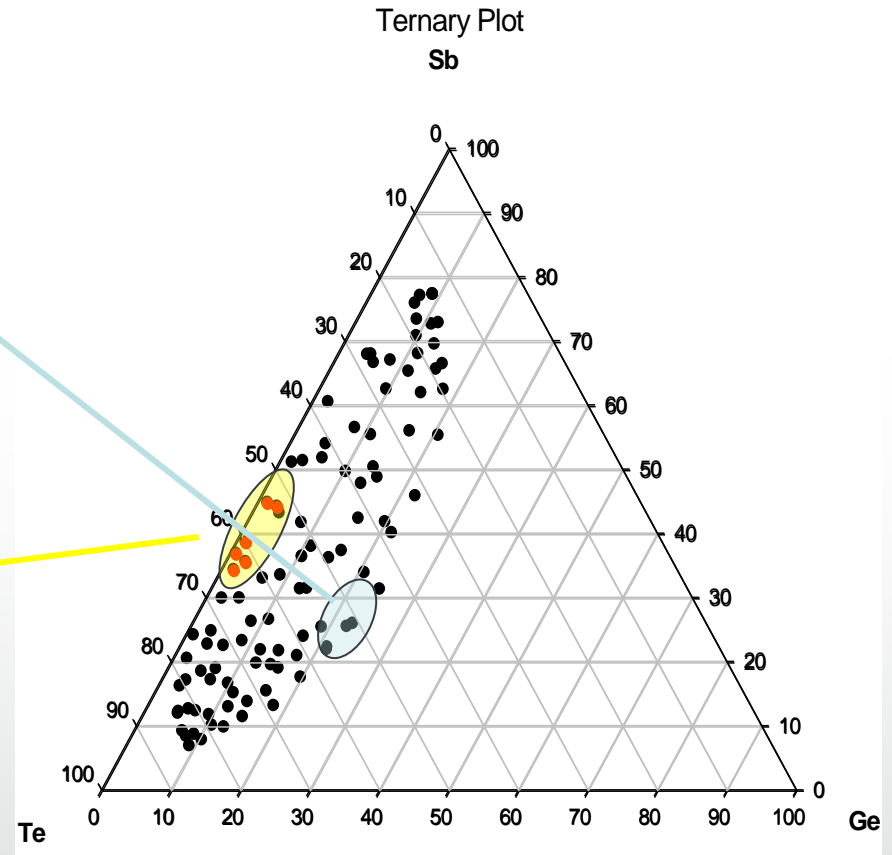
Optical Screening

Hot plate heating

25 C

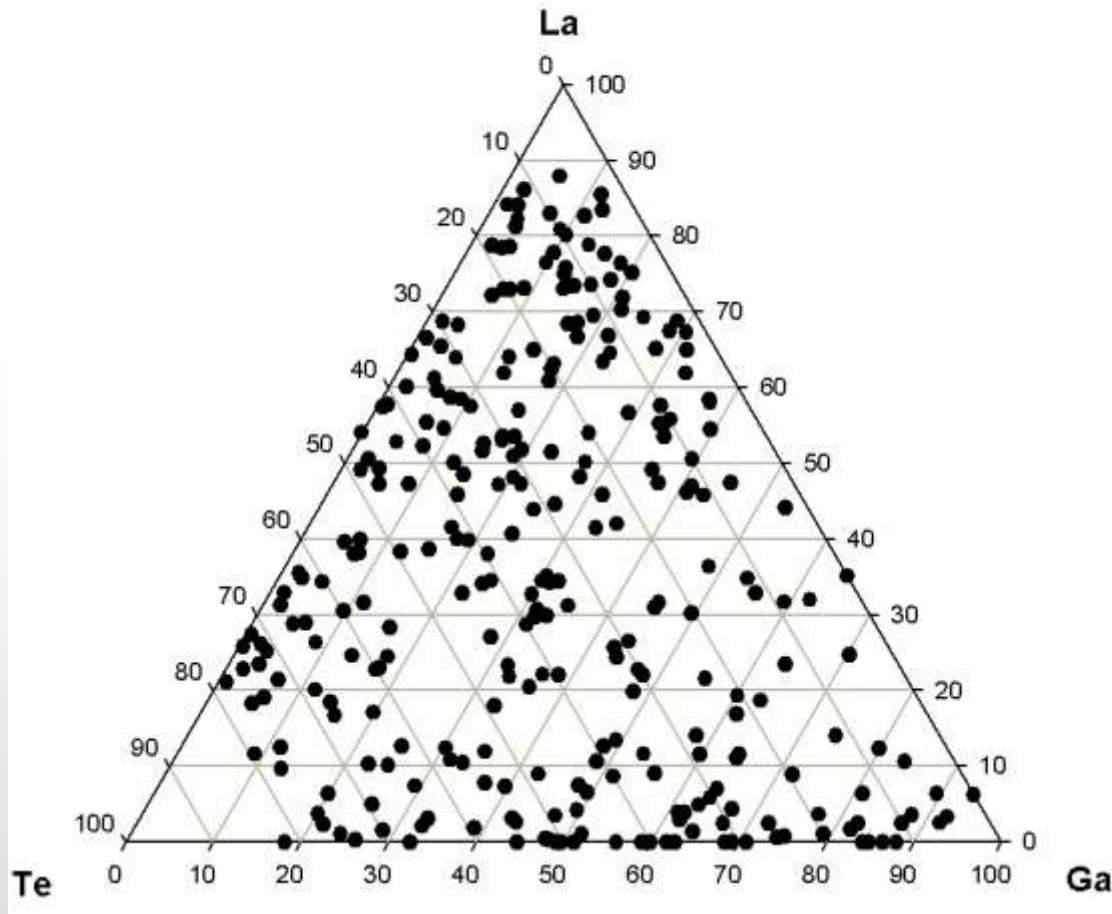


High Sensitivity B&W Progressive Scan
1392 x 1040 pixel CCD Camera
Starlight Express Ltd. Model SXV-H9



Full Compositional Analysis of Ga:Te:La System

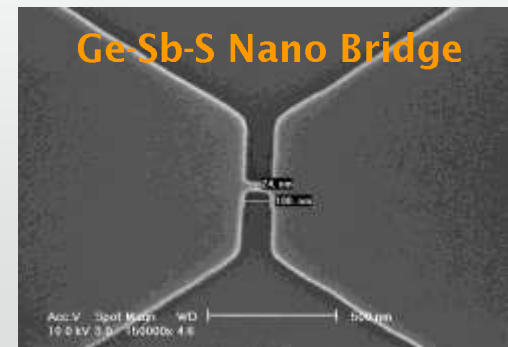
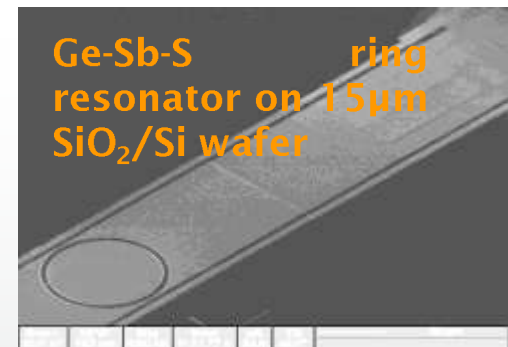
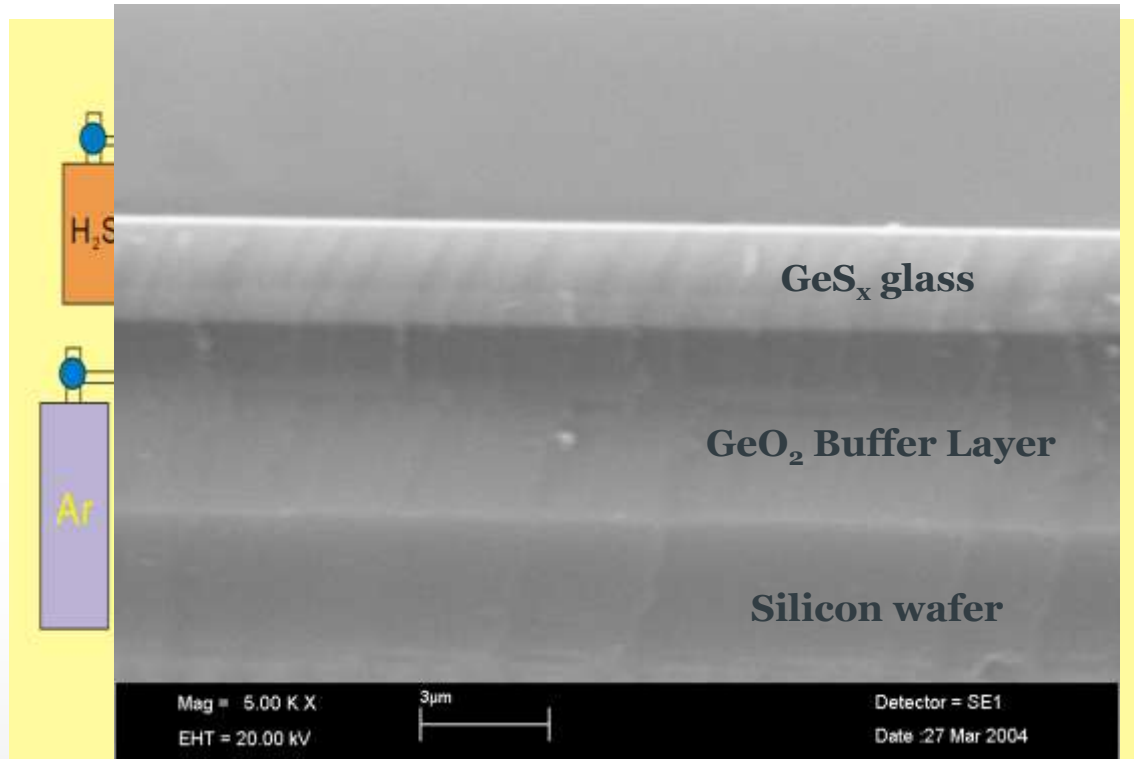
EDS 1613 + 1659



Optical Fiber



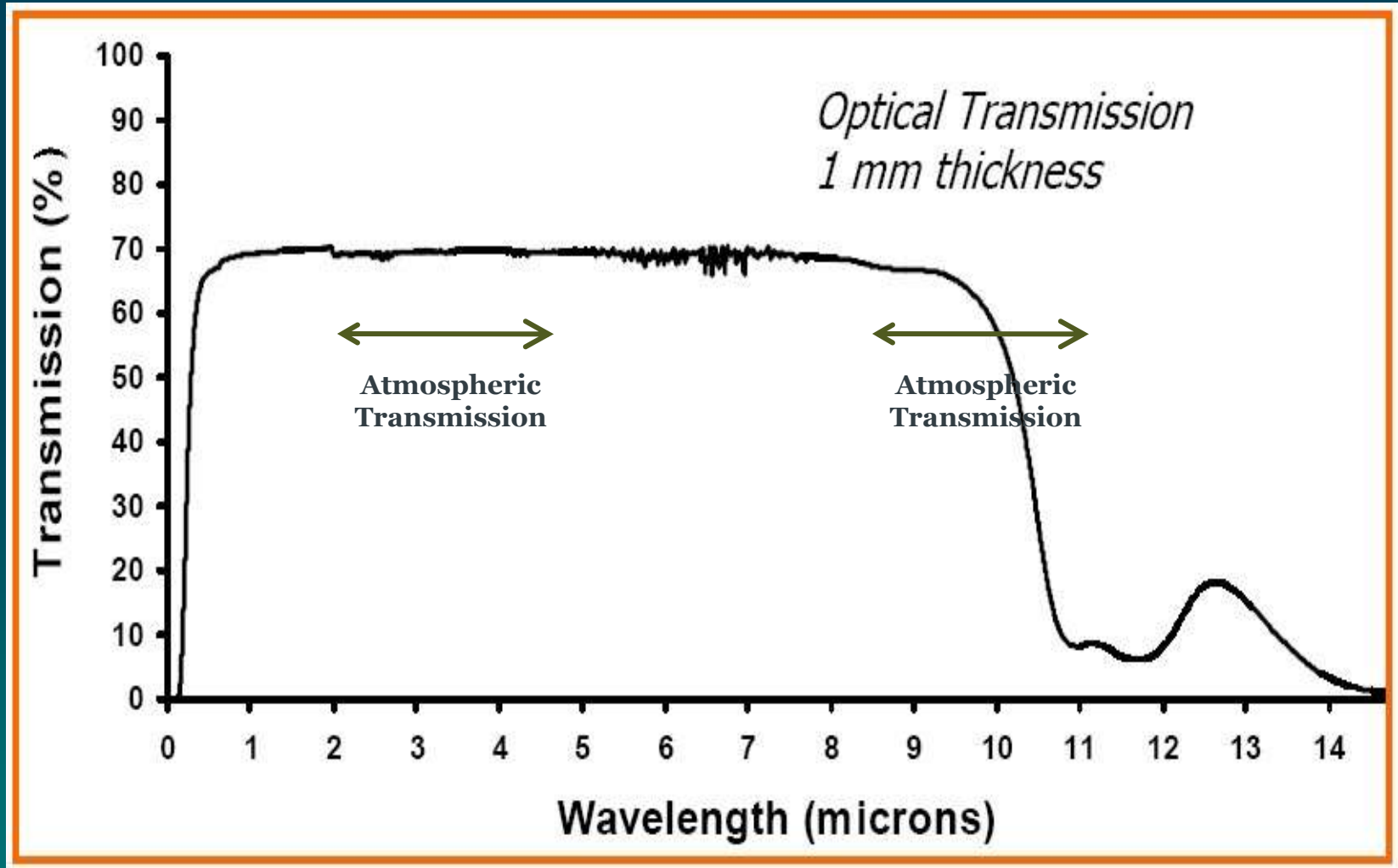
Chemical Vapour Deposition



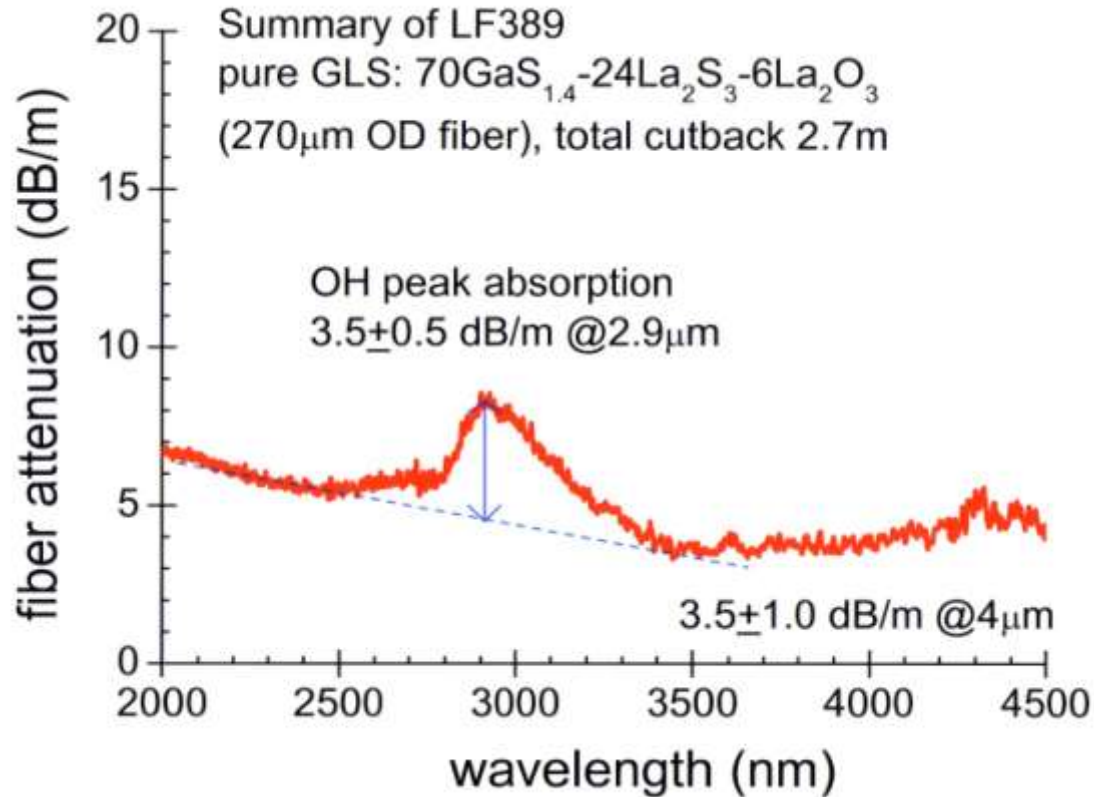
Impurities in glasses prepared by different methods (data in ppm)

Glass	Preparation Method	Cr	Fe	Co	Ni	Cu	Zn
Ge:S	CVD	<0.005	<0.0 5	<0.0 5	<0.0 5	<0.0 5	<0.05
Ga:La:S	Melt Quench	0.02	0.06	nil	0.04	0.1	<0.5

Optical Transmission



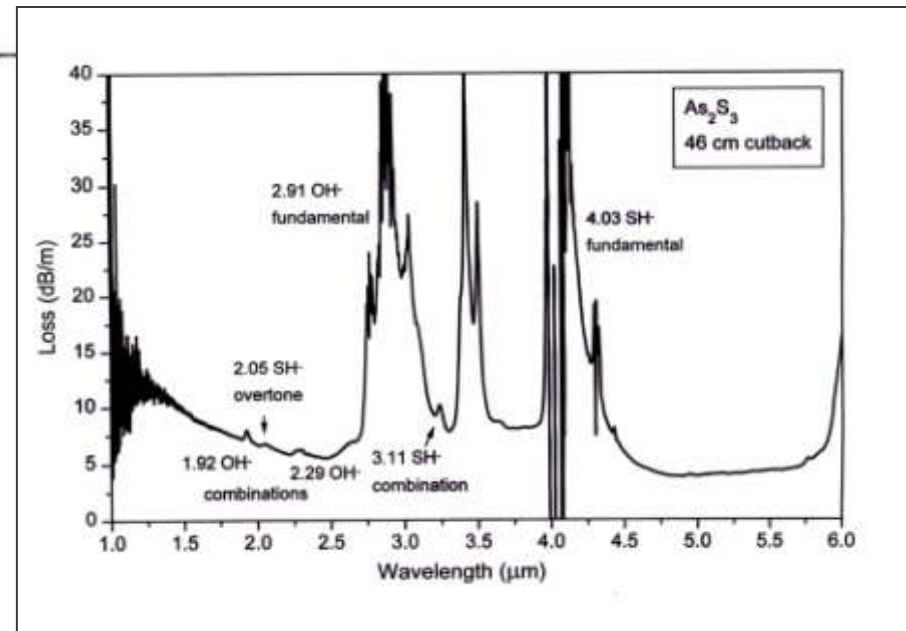
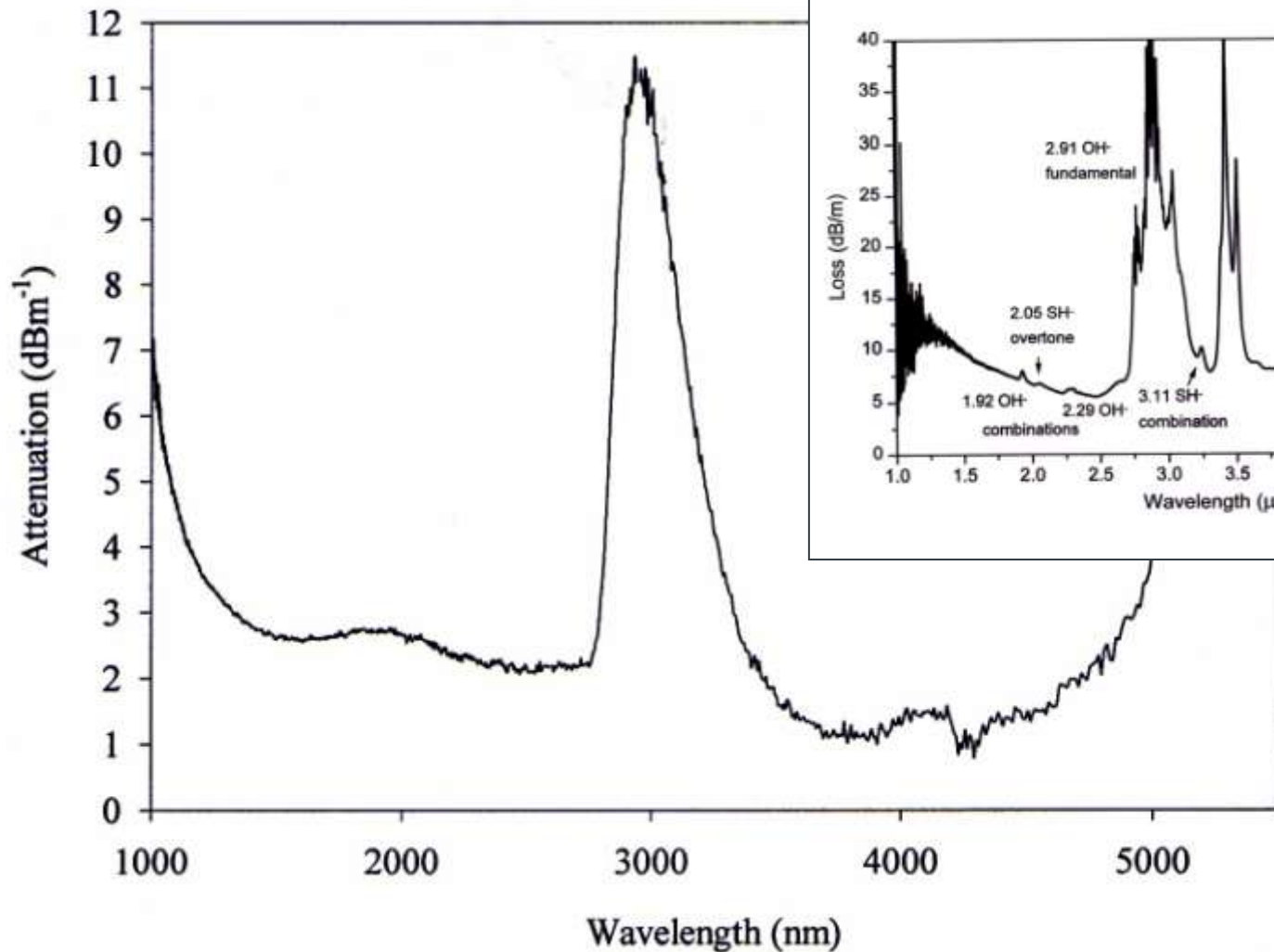
Optical Fibre Results



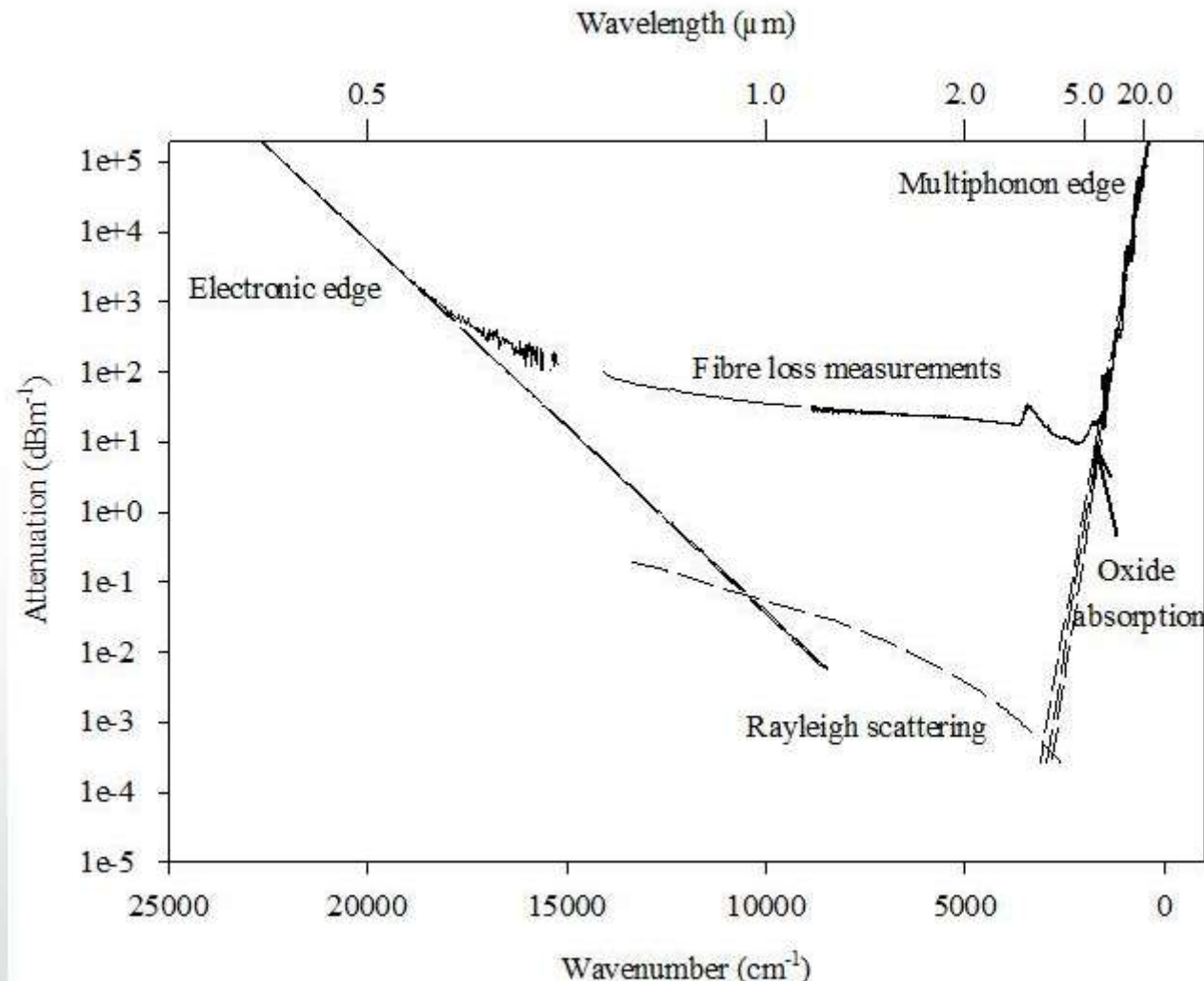
g

	Internal Loss	Optical Fibre
1992	0.1 cm^{-1}	40 – 60 dB/m
2002	0.02 cm^{-1}	2-3 dB/m
Target*	0.004	< 0.2 dB/m

Comparison of GLS fibre with Commercial Arsenic-based fibre



Theoretical Loss Limits



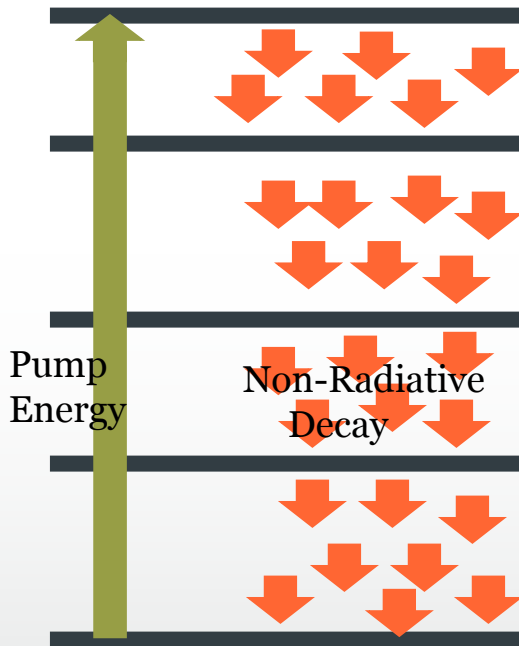
Results in practice are far from predictions!

Active Devices

- Mid-IR sources
- Microspheres
- Emerging technologies

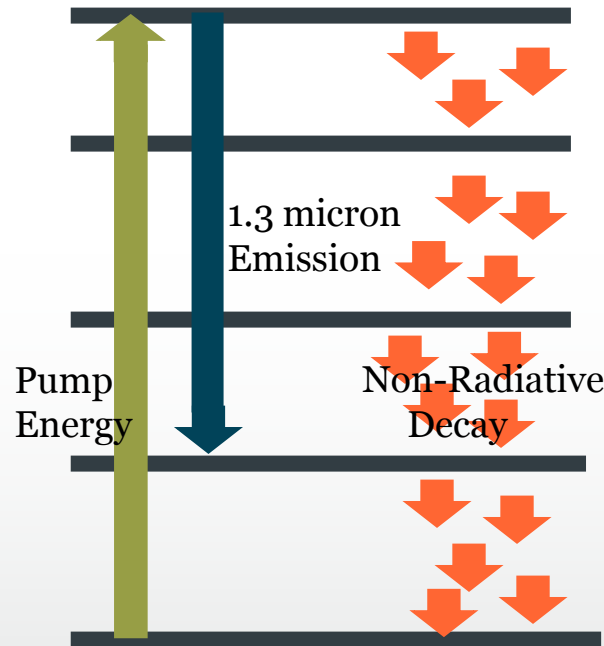
Glass Structure & Multiphonon Decay

Pr^{3+}



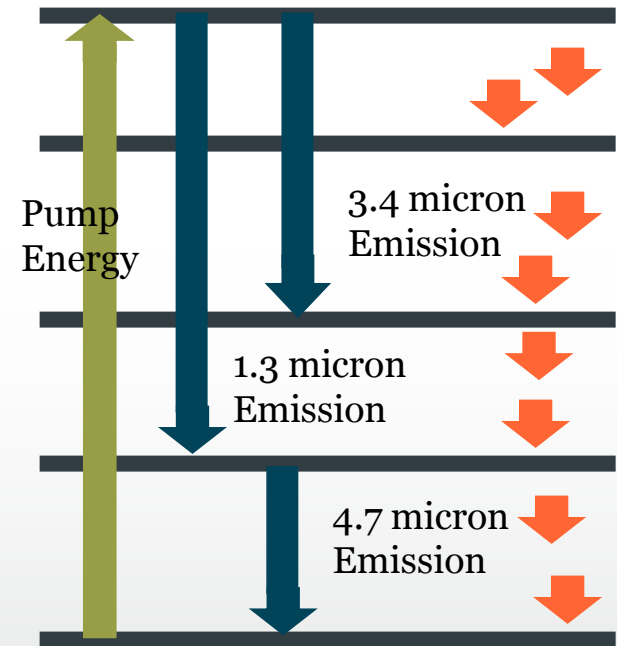
Oxide Glass
High Phonon Energy
Efficiency = 0%

Pr^{3+}



Fluoride Glass
Intermediate Phonon Energy
Efficiency = 6%

Pr^{3+}



Chalcogenide Glass
Low Phonon Energy
Efficiency = 60%

Rare Earth Doping

Rare Earth Elements														Y 39				
La 57	Ce 58	Pr 59	Nd 60	Pm 61	Sm 62	Eu 63	Gd 64	Tb 65	Dy 66	Ho 67	Er 68	Tm 69	Yb 70	Lu 71				
Lanthanides																		
H																	He	
Li	Be											B	C	N	O	F	Ne	
Na	Mg											Al	Si	P	S	Cl	Ar	
K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr	
Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe	
Cs	Ba		Lu	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rn
Fr	Ra	An	Lr															

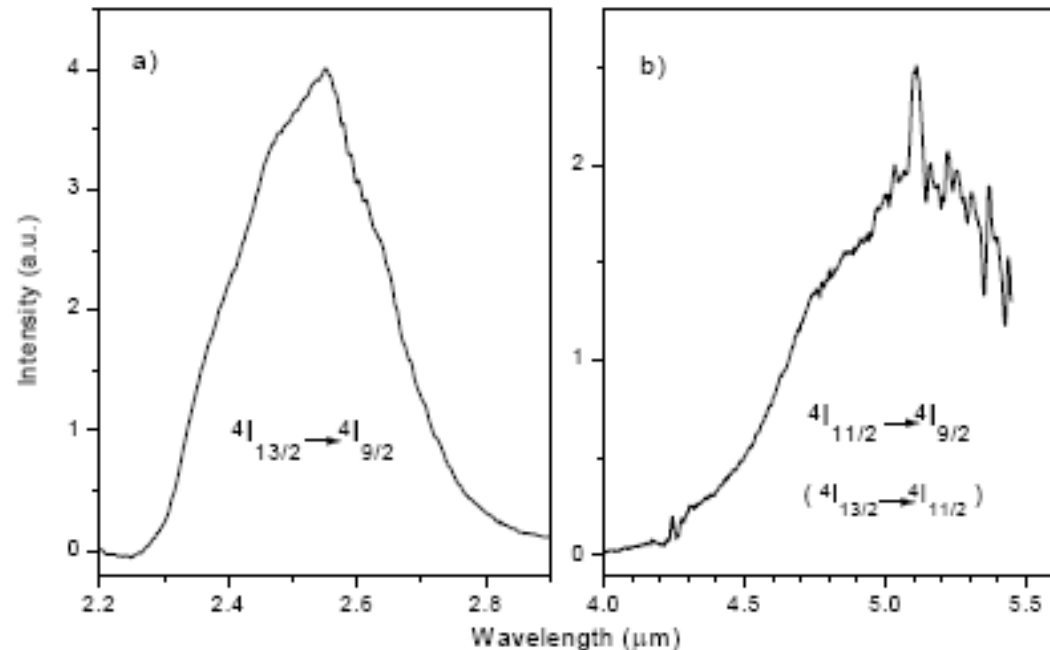


Fig. 6 Emission from the two lowest Nd³⁺ levels, $4I_{13/2}$ (a) and $4I_{11/2}$ (b), in 1.5 mol% Nd₂S₃ doped GLS glass pumped at 815 nm with a Ti:sapphire laser and with a 300-mm monochromator and a liquid nitrogen cooled InSb measured detector

Nd³⁺

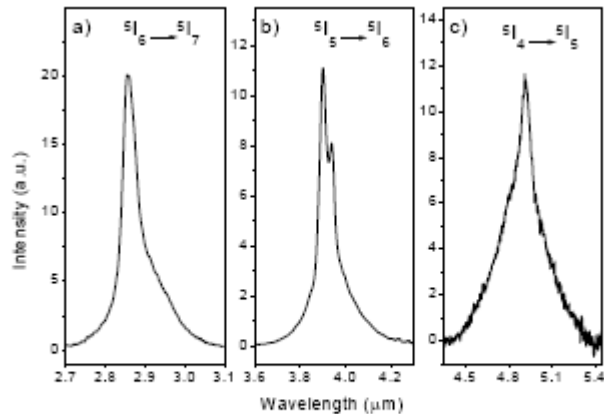


Fig. 5 Mid-infrared emission a) from the 5I_6 level, b) from the 5I_5 level, and c) from the 5I_4 level in Ho(1.5%):GLS pumped at 0.76 μm

Ho³⁺

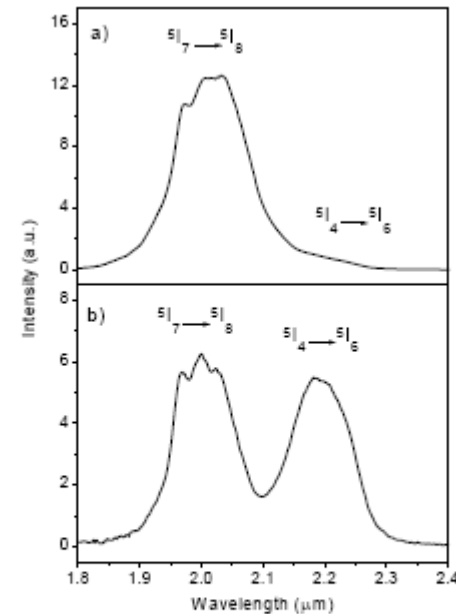


Fig. 3 Overlapping emission at 2.0 and 2.2 μm from the 5I_7 and the 5I_4 levels in Ho(1.5%):GLS pumped at 0.76 μm
a) continuous wave pump laser
b) pump laser chopped at 400 Hz

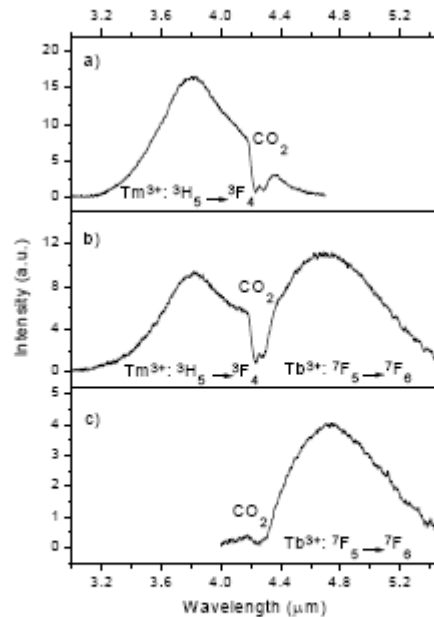


Fig. 7 Uncorrected fluorescence spectra of
a) Tm(1.5%):GLS glass pumped at 0.7 μm
b) Tm(1.5%),Tb(0.2%):GLS glass pumped at 0.7 μm
c) Tm(1.5%),Tb(0.2%):GLS glass pumped at 2 μm

Tm³⁺ and Er³⁺

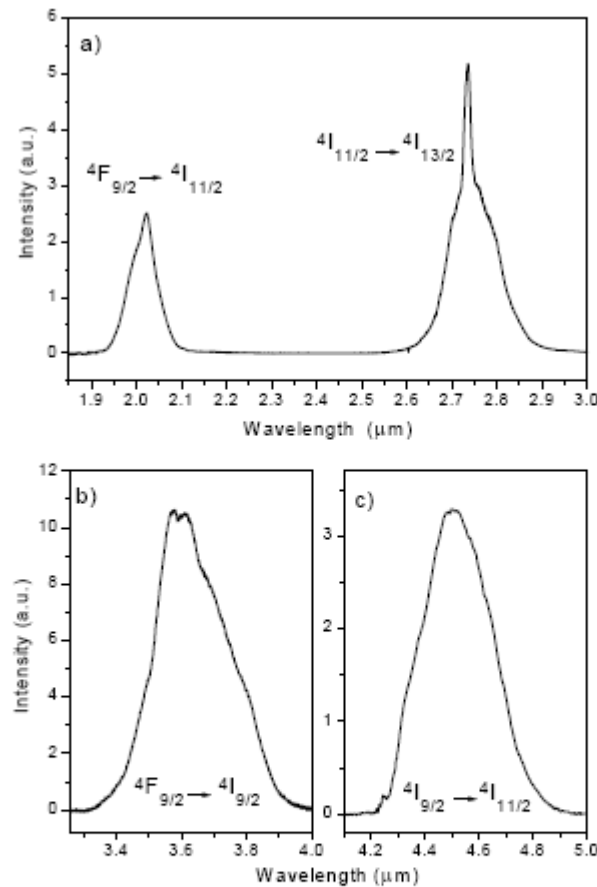


Fig. 4 Fluorescence spectra of 1.57 mol% Er³⁺ doped GLS glasses and fibres
a) 2.0 μm and 2.75 μm emission from 3.9 cm of 270 μm diameter fibre pumped with 60 mW at 660 nm
b) 3.6 μm emission from 8.6 cm of 270 μm diameter fibre pumped with 70 mW at 660 nm
c) 4.5 μm emission from a bulk glass sample pumped with 570 mW at 810 nm

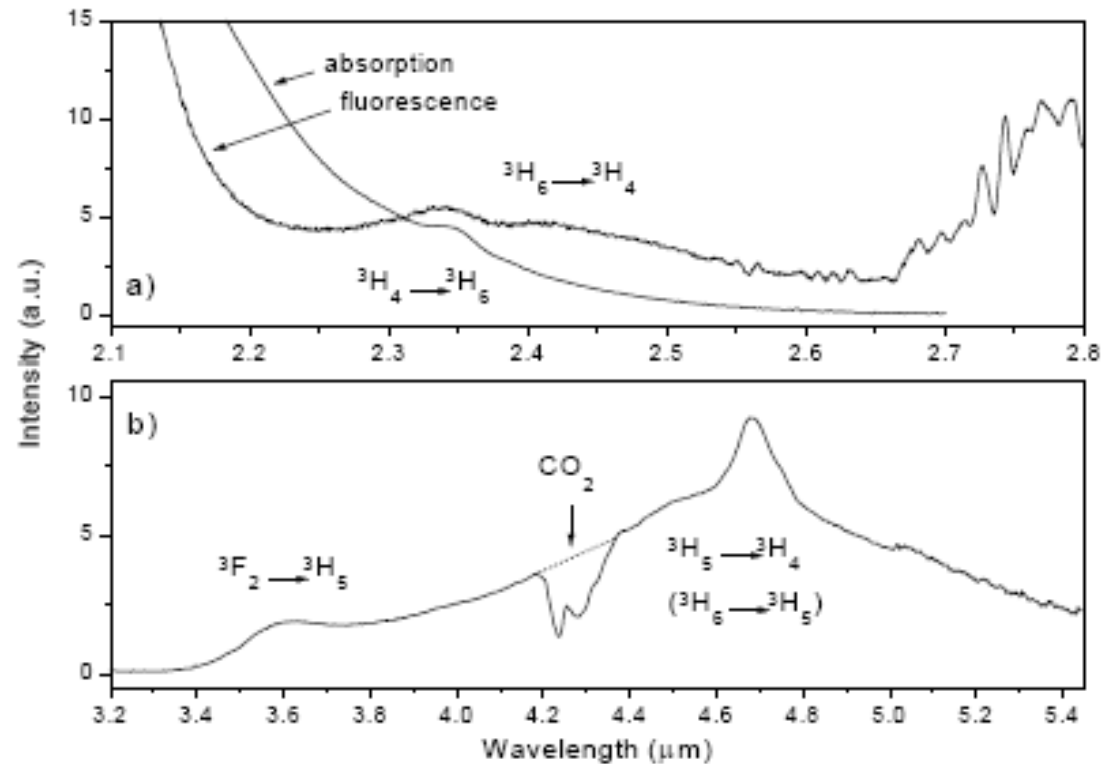


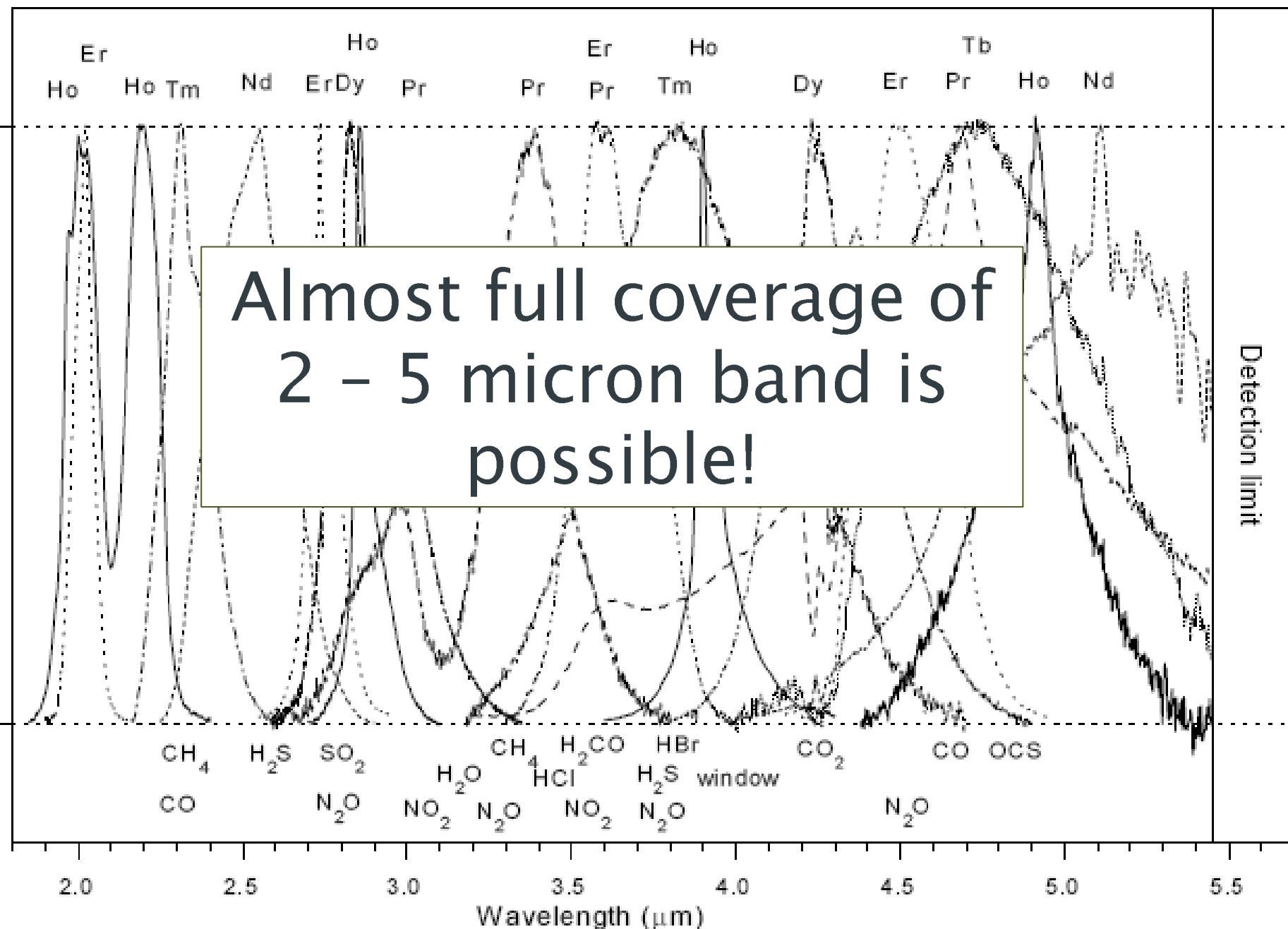
Fig. 3.4.4 Emission from three lower Pr^{3+} levels:
a) Fluorescence from 2 m long, 500 ppm Pr^{3+} doped GLS fibre pumped with 1.064 μm Nd:YAG laser and absorption spectrum
b) Fluorescence from 2000 ppm Pr^{3+} doped GLS bulk glass pumped with 2 μm Tm:YAG laser

Pr^{3+}

Normalised intensity

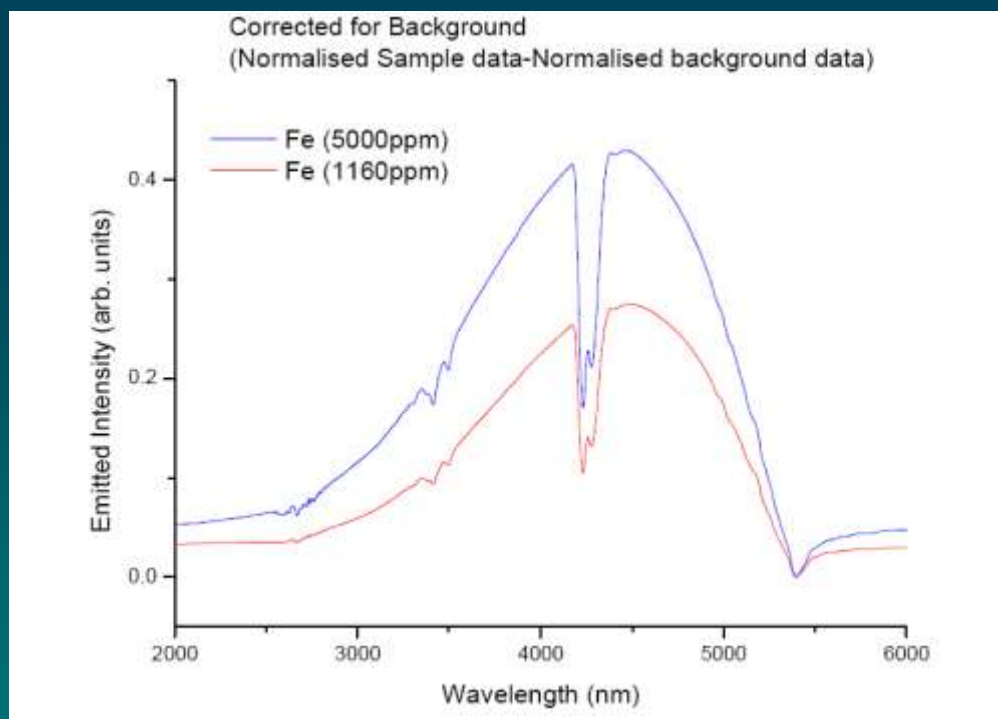
Detection limit

Almost full coverage of
2 – 5 micron band is
possible!



Transition Metal Doping

IIIB	IVB	VB	VIB	VII B	—— VII ——			IB	IB
21	22	23	24	25	26	27	28	29	30
Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn



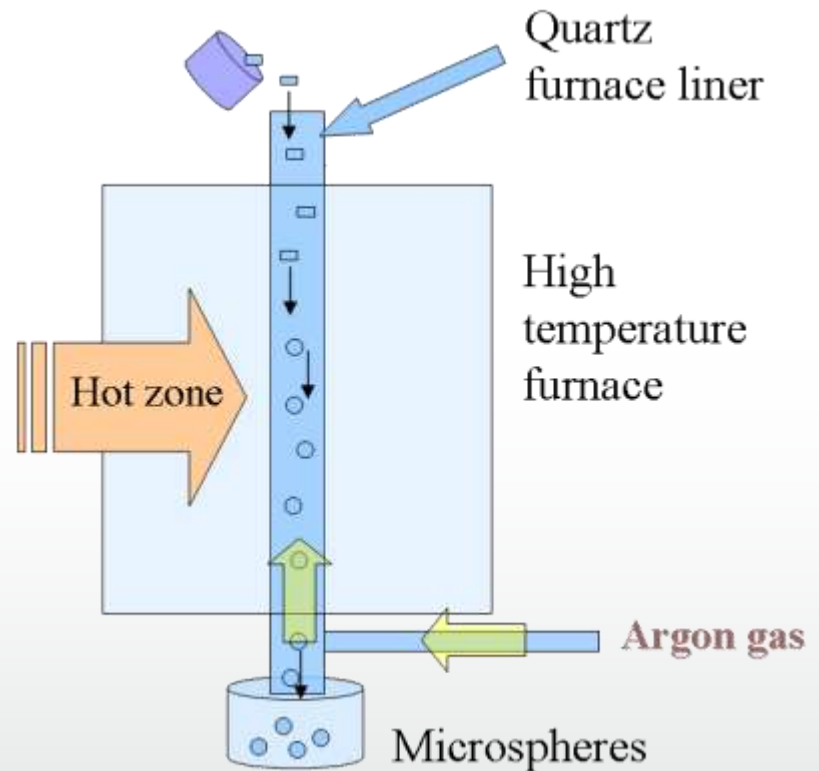
Direct Emission Mid-IR Lasers

Research Contract from UK Laser Coalition
DSTL, Qinetiq, BAe Systems, Selex

Microspheres

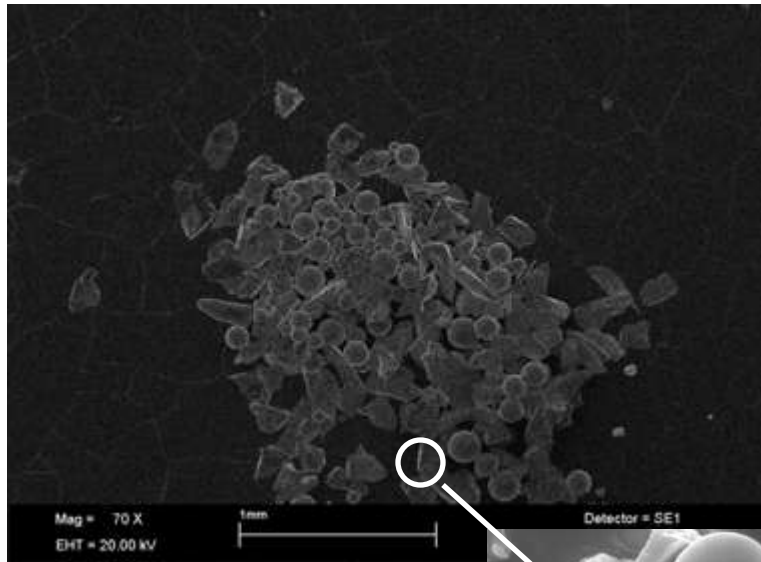


Microsphere Fabrication

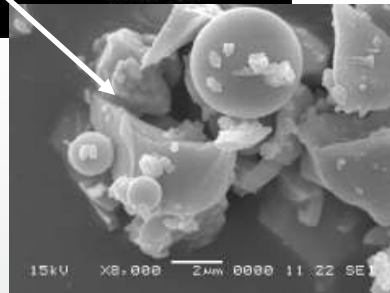


Microsphere diameters 500 nm to 500 μm

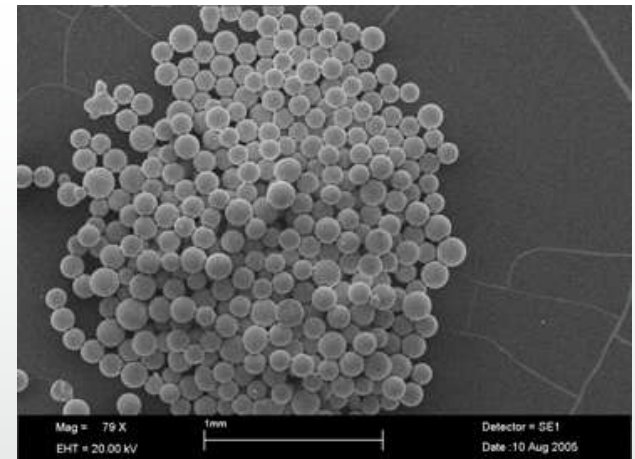
Sphere Sorting



as fabricated...



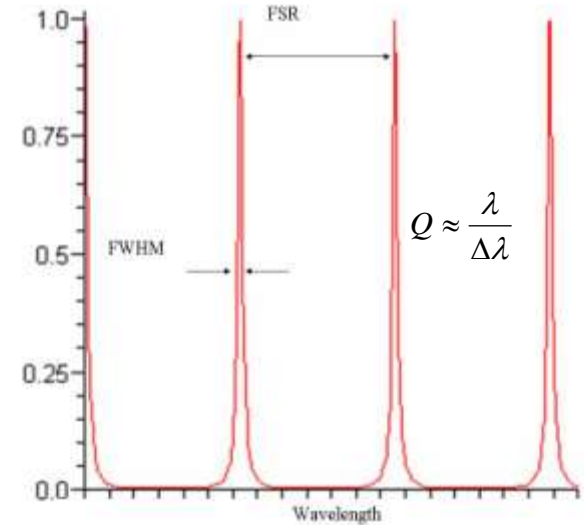
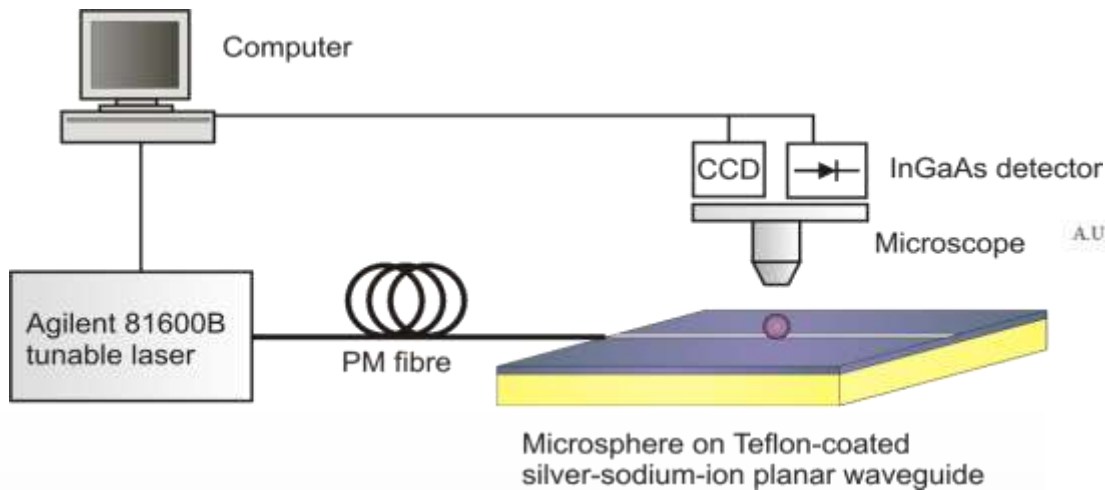
sieving
sedimentation
rolling



Size range:
100's nm to 100's microns

Greg Elliott

Microsphere Characterization

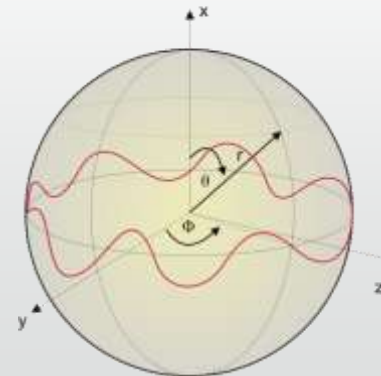


Quality Factor: Q where $1/Q = 1/Q_{\text{material}} + 1/Q_{\text{surface}} + 1/Q_{\text{curvature}} + 1/Q_{\text{coupling}}$

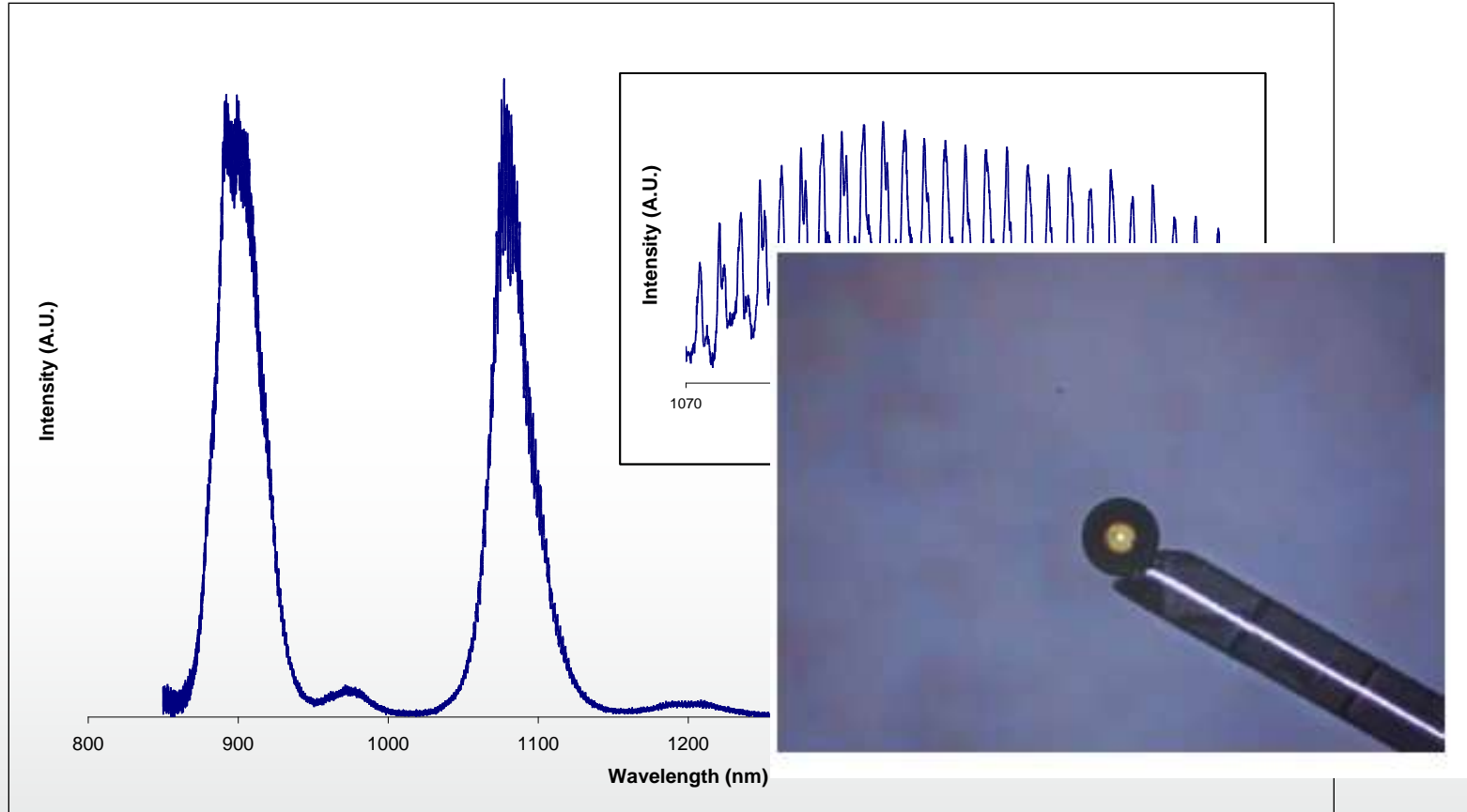
$$Q_{\text{predicted}} = \sim 4 \times 10^9$$

$$Q_{\text{measured}} = 8 \times 10^4$$

Gregor R. Elliott, Daniel W. Hewak, G. S. Murugan, and James S. Wilkinson, "Chalcogenide glass microspheres; their production, characterization and potential", Optics Express, Vol. 15, Issue 26, pp. 17542-17553

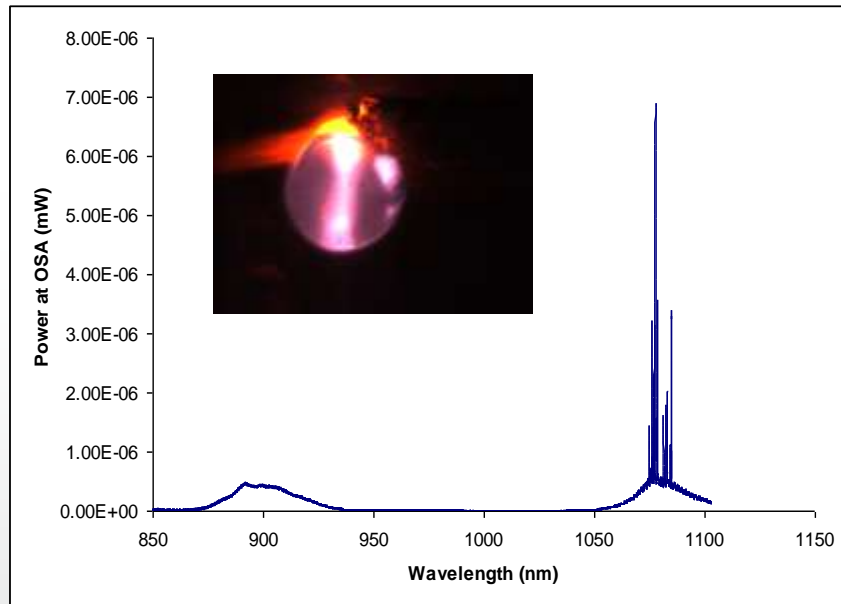


Improved Coupling

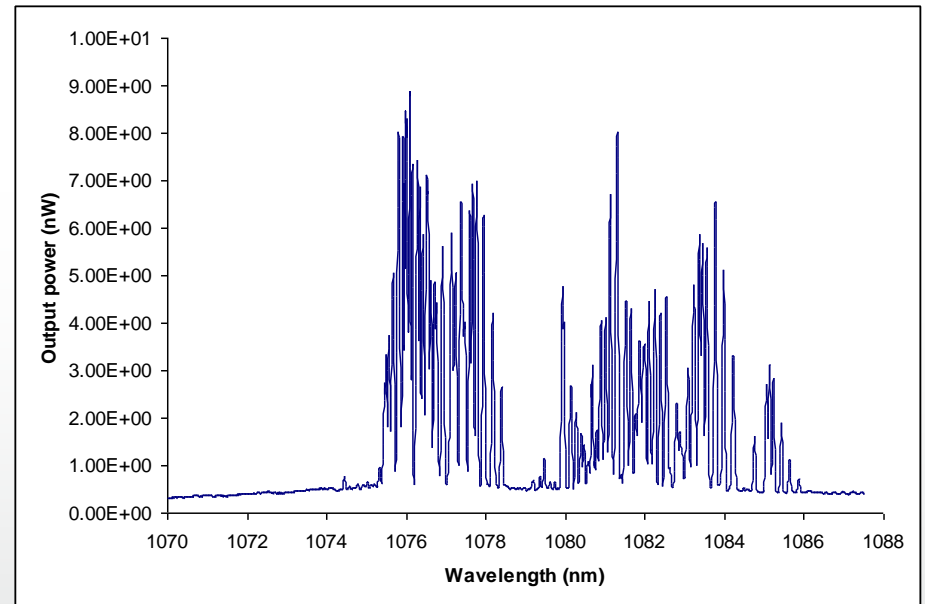


Fluorescent spectrum from GLS microsphere doped with 1.5mol% Nd. Inset, close up on WGM. This measurement was taken using a fibre coupled microsphere.

Initial Lasing Observations



217 mW Pump Power



Maximum Pump Power

Laser Threshold: 82 mW delivered to sphere

(19) **United States**

(12) **Patent Application Publication**
ABDELDAYEM

(10) **Pub. No.: US 2008/0212620 A1**

(43) **Pub. Date: Sep. 4, 2008**

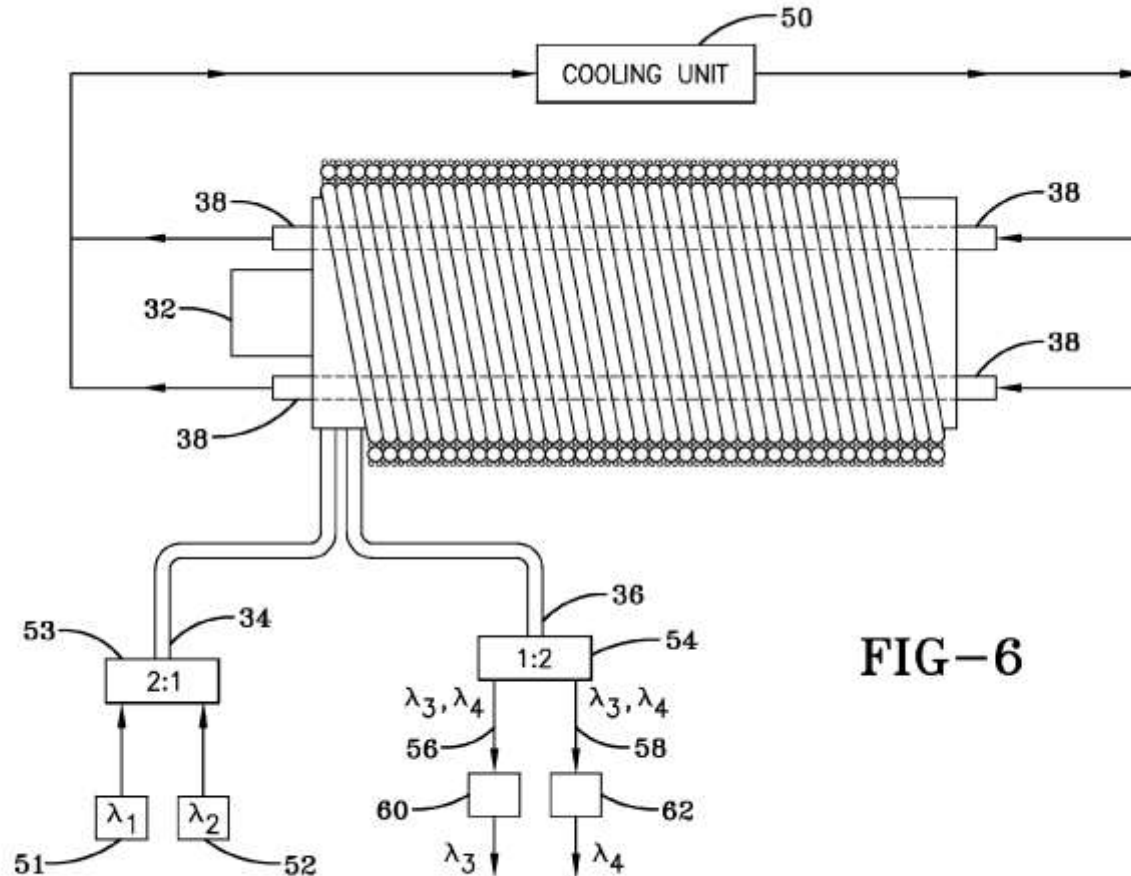
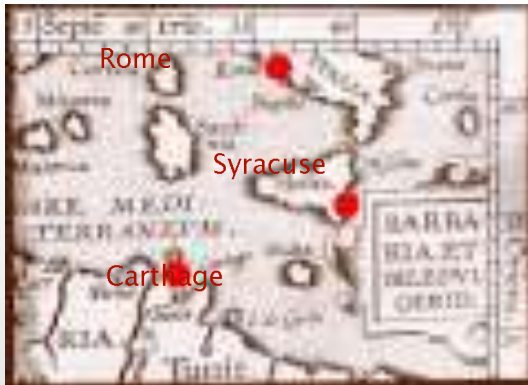


FIG-6

UV-Vis-NIR Spatial Beam Combination

First ever defence application ...



Siege of Syracuse
2nd Punic War - 214 BC



Burning Mirrors of Archimedes

Characteristics of Microsphere Lasers

- Extremely low threshold
- Simple, robust cavity, easily fabricated
- Integratable with planar or fibre technology
- Potential for new wavelengths in IR

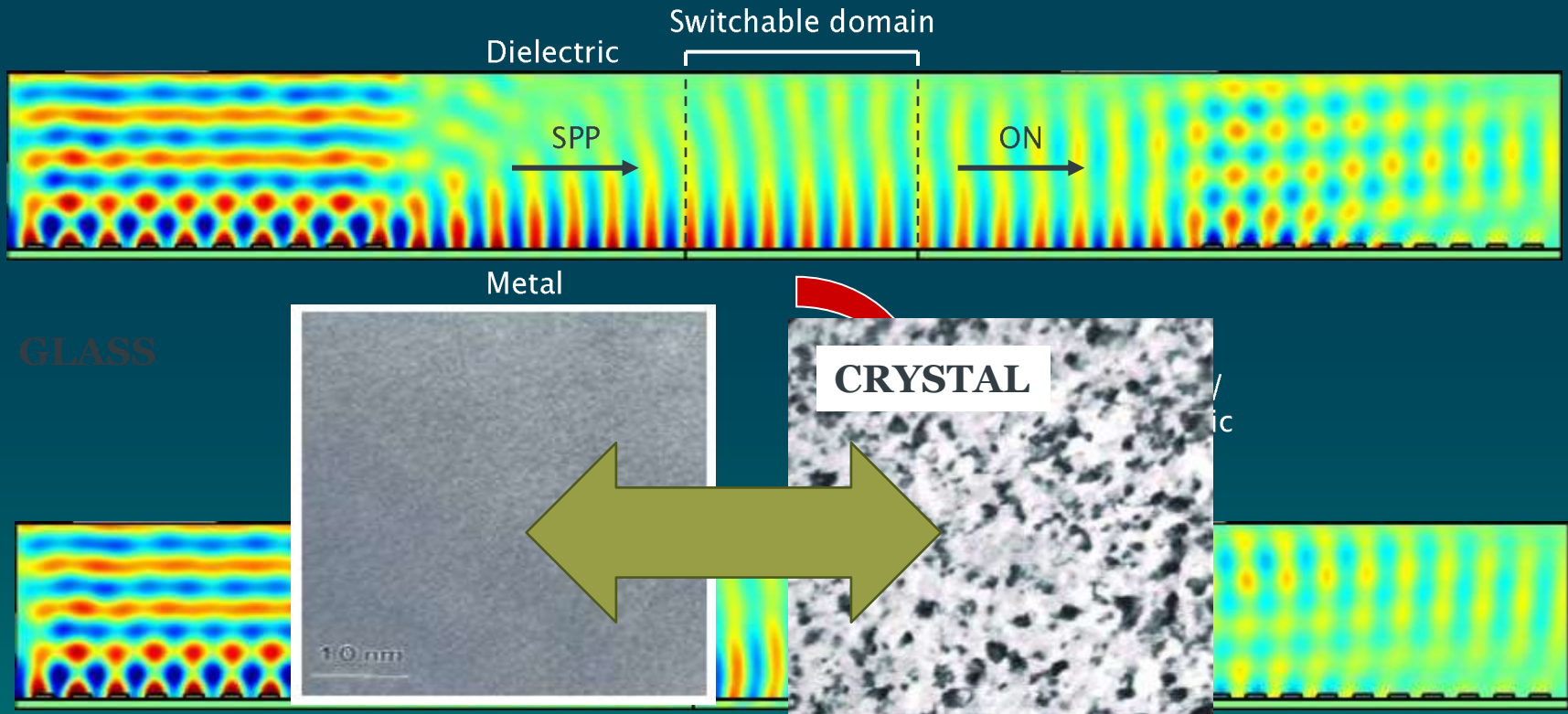
Application of Phase Change in Nanophotonics



- Plasmonics
- Metamaterials

<http://www.nanophotonics.org.uk/>

Active Plasmonics ... The Concept



- Transient changes in waveguide properties control SPP propagation.
- Short-range atomic transitions
 - Low free electron density
 - High activation energy
 - High resistivity
 - Transparent
 - High free electron density
 - Low activation energy
 - Low resistivity
 - Opaque
- APL 84, 1416 (2004)

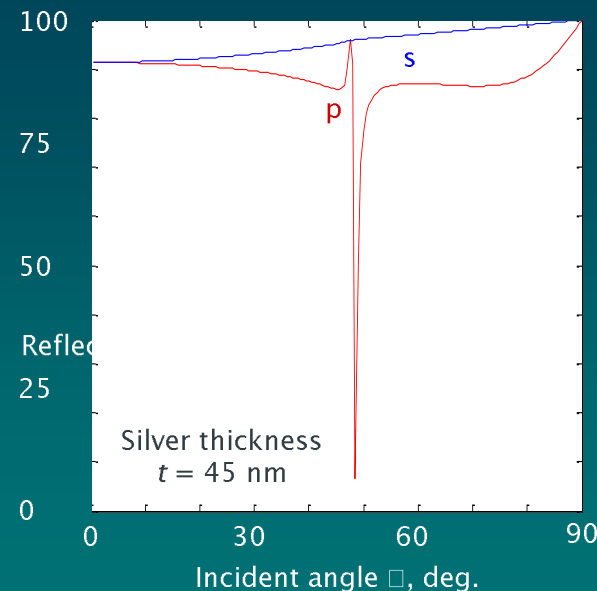
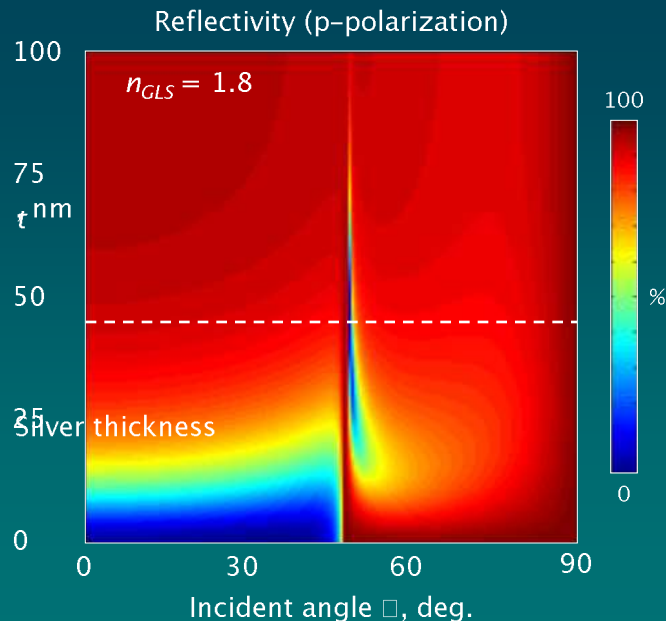
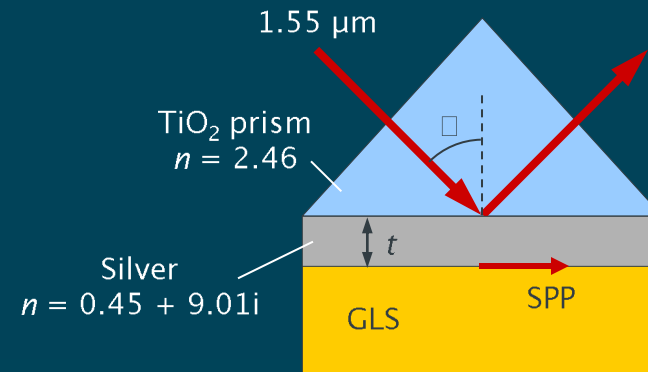
UNIVERSITY OF
Southampton

Ge-Sb-Te (GST) unsuitable for plasmonics

- high index
- highly absorbing)

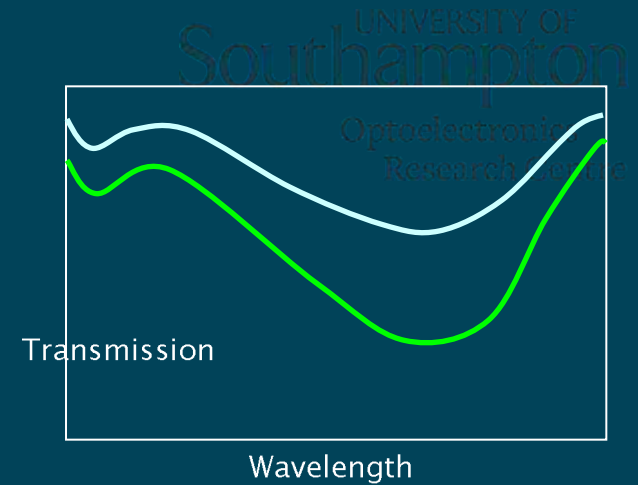
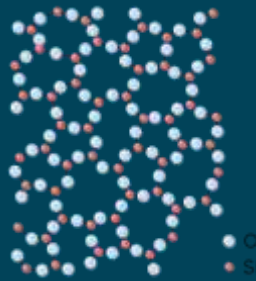
Ga-La-S is much better:

- transparent above ~ 700 nm
- High damage threshold
- Easily polished and coated
- non-toxic

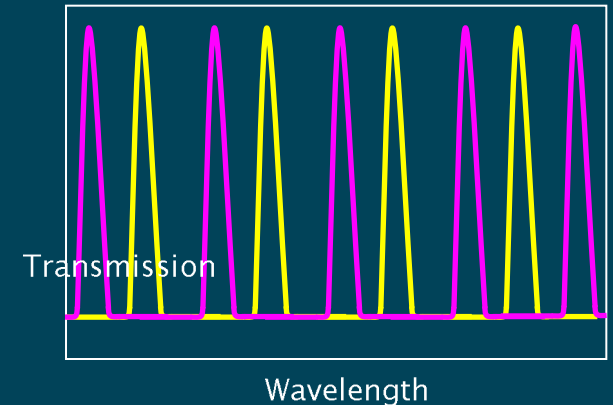
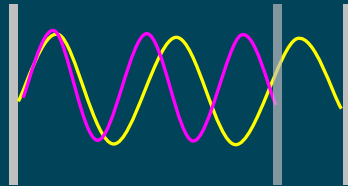


Resonance switching

Nonlinear Medium

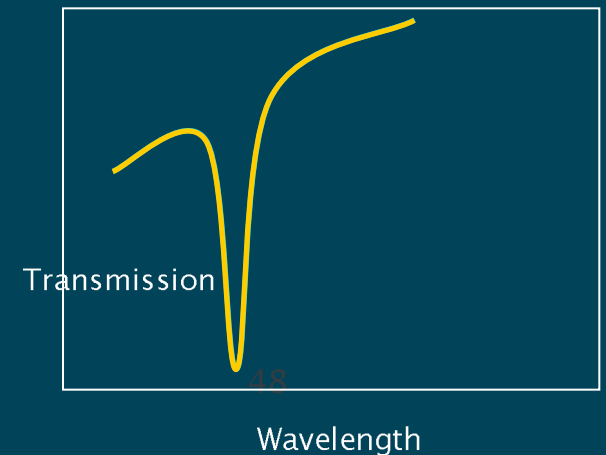
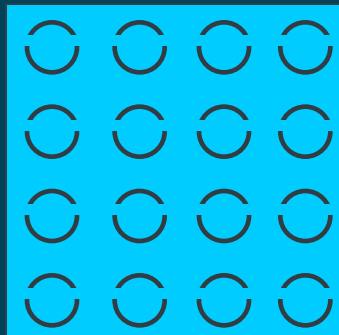


Fabry-Perot Resonator



Can we switch our metamaterials?

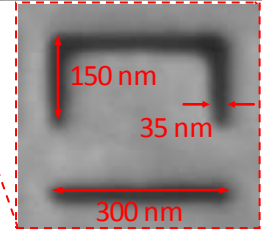
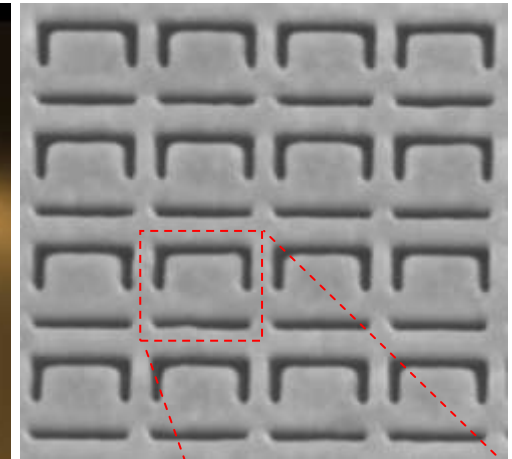
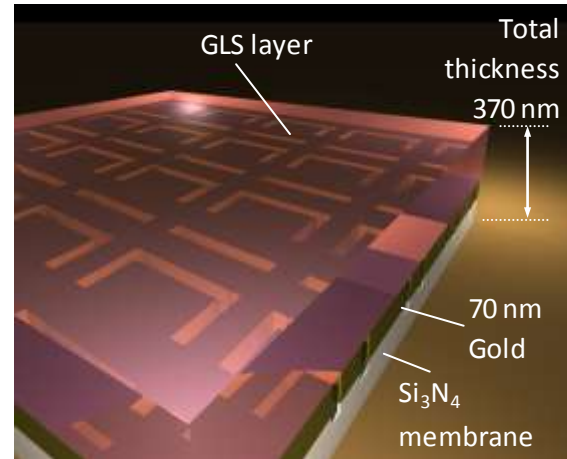
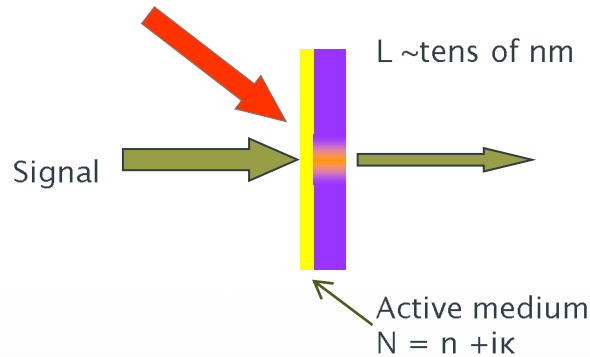
Planar Metamaterial



Chalcogenide metamaterial hybrid

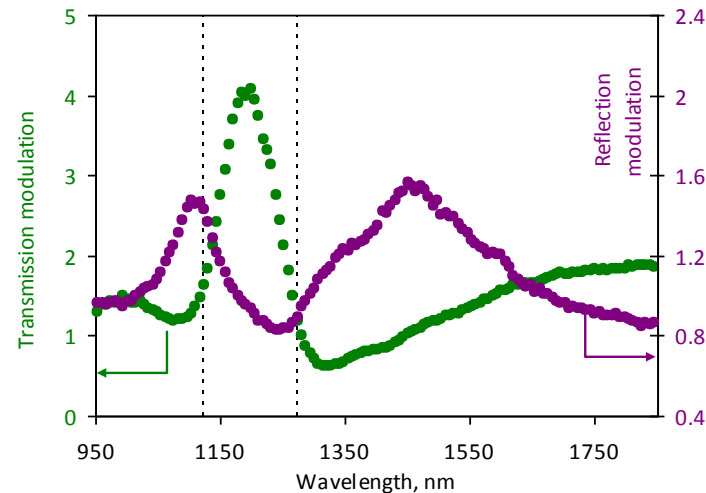
Electro-optic modulator

Control
(Induces change Δn and/or Δk)



Gallium Lanthanum Sulphide:

- Optically/electrically-induced threshold switching: amorphous – crystalline
- Transmission contrast 4:1 in a device only $1/3$ of a wavelength thick.
- Operational band tuneable by design across VIS-IR range



Concluding Remarks

- Purify, purify, purify
- Don't (always) blame the composition
- Consider other geometries, emerging technologies
- Collaborate openly* (or find a very wealthy sponsor)
- Learn from history



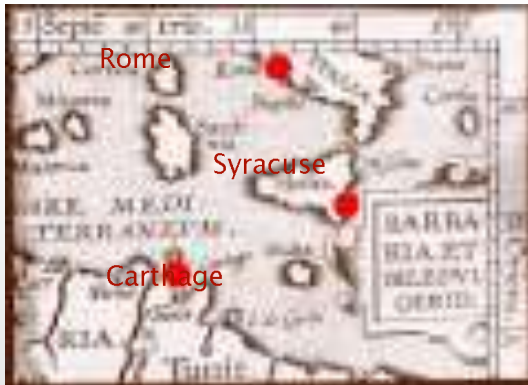
Four years later....

*when you can



UV-Vis-NIR Spatial Beam Combination

First ever defence application ...



Siege of Syracuse
2nd Punic War - 214 BC



Burning Mirrors of Archimedes